#### Chapter 7 Study Guide ANSWERS

#### Practice Problems:

Name the following compounds:

- BeO Beryllium oxide
- Cs<sub>8</sub>N Cestium nitride
- 3. RbCl Rubidium chloride

Determine the formula of the following compounds:

- Aluminum oxide Al<sub>2</sub>O<sub>3</sub>
- 2. Calcium bromide CaBr:
- Lithium sulfide LigS
- 4. Barium fluoride HaF:
- Aluminum chloride AlC1;

#### Practice Problems:

Name the following compounds:

- CrF<sub>2</sub> Chromium (II) fluoride
- 2. NiO Nickel (II) oxide
- Fe<sub>2</sub>O<sub>3</sub> Iron (III) oxide
- 4. VCI Vanadium (I) chloride
- SnO<sub>2</sub> Tin (IV) oxide

Determine the formula of the following compounds:

- 1. Iron (III) sulfide Fe Si
- 2. Lead (II) oxide PbO
- Chromium (III) iodide Crl<sub>1</sub>
- 4. Copper (I) oxide CusO
- Cobalt (II) nitride CoiN:

#### Practice Problems:

Name the following compounds:

- Al(NO<sub>3</sub>)<sub>3</sub> Aluminum nitrate
- KMnO<sub>4</sub> Potassium permanganase
- FeSO<sub>3</sub> Iron (II) sulfite
- 4. (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> Ammonium phosphate
- NaCN Sodium cyanide.

Determine the formula of the following compounds:

- Copper (II) nitrite Cu(NO<sub>2</sub>)<sub>2</sub>
- Ammonium oxide (NH<sub>0</sub>)<sub>2</sub>O
- 3. Sodium hydroxide NaOH
- Iron (III) chromate Fe<sub>2</sub>(CrO<sub>4</sub>):
- Calcium carbonate CaCO<sub>3</sub>

# **Chemistry Study Guide Answer**

Susan McMurry, John McMurry

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