4

Atomic Structure

Reviewsheet

A. Completion

Use this completion exercise to check your understooding of the concepts and terms that are introduced in this chapter. Each black can be completed with a term, shore phoses, or number.

It was ___7__ who discovered the nucleus of the atom. The nucleus has a ___8__ charge and it occupies a very small volume of the atom. In contrast, the negatively charged ___9__ occupy most of the volume of the atom.

The number of __19_ in the nucleus of the atom is the atomic __11_ of that element: Because atoms are electrically neutral, the number of protons and __12_ in an atom are equal. The sum of the __13_ and neutrons is the mass number. Atoms of the same element are identical in most respects, but they can differ in the number of __14_ in the nucleus. Atoms that have the same number of protons but different mass numbers are called __15__.

The ___16__ of an element is the weighted average of the __2 masses of the isotopes of that element. Two isotopes of sulfur are __12 S and __14 S. An atom of the sulfur—32 isotope contains __17__ protons and __18__ neutrons. The sulfur—34 isotope has __19__ protons and __20__ neutrons.

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sob aboute	4.2
repative	4.2
6x01003	4.2
1360 HOWE	4.2
nucleus .	4.9
Rutnertore	6.5
positive	
electrans	4.8
prodons (elec)	14
ramber	4.4
evectore	4.4
newtrons	
isotopes	4.6
16	4.5
	4.6
	4.6
18	4.5
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- 1. Understanding the eBook Chemistry Atomic Structure Chapter Review Answers
 - The Rise of Digital Reading Chemistry Atomic Structure Chapter Review Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry Atomic Structure Chapter Review Answers
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Atomic Structure Chapter Review Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry Atomic Structure Chapter Review Answers
 - Personalized Recommendations
 - Chemistry Atomic Structure Chapter Review Answers User Reviews and Ratings
 - Chemistry Atomic Structure Chapter Review Answers and Bestseller Lists
- 5. Accessing Chemistry Atomic Structure Chapter Review Answers Free and Paid eBooks
 - Chemistry Atomic Structure Chapter Review Answers Public Domain eBooks
 - Chemistry Atomic Structure Chapter Review Answers eBook Subscription Services
 - Chemistry Atomic Structure Chapter Review Answers Budget-Friendly Options

- 6. Navigating Chemistry Atomic Structure Chapter Review Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Chemistry Atomic Structure Chapter Review Answers Compatibility with Devices
 - Chemistry Atomic Structure Chapter Review Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Atomic Structure Chapter Review Answers
 - Highlighting and Note-Taking Chemistry Atomic Structure Chapter Review Answers
 - Interactive Elements Chemistry Atomic Structure Chapter Review Answers
- 8. Staying Engaged with Chemistry Atomic Structure Chapter Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Atomic Structure Chapter Review Answers
- 9. Balancing eBooks and Physical Books Chemistry Atomic Structure Chapter Review Answers
 - $\circ\,$ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Atomic Structure Chapter Review Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Atomic Structure Chapter Review Answers
 - Setting Reading Goals Chemistry Atomic Structure Chapter Review Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Atomic Structure Chapter Review Answers
 - Fact-Checking eBook Content of Chemistry Atomic Structure Chapter Review Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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