

## Chapter 5 Review Worksheets Answers

### Section 1:

1. C
2. D
3. B
4. C
5. D
6. C
7. Fluorine isotope with mass number 19 and atomic number 9:
  - a. 9
  - b. 10
  - c.  $^{19}\text{F}$
8. Samarium (Sm) is a lanthanide, what is the element below Sm on the periodic table? Difference in atomic #s?
  - a. Plutonium Pu
  - b. 32
9. An isotope contains 53 protons, 78 neutrons, & 54 electrons
  - a. 53
  - b. 131
  - c. Iodine
  - d. Fluorine (F), Chlorine (Cl), Bromine (Br), & Astatine (At)
10. Group = Vertical column      Period = Horizontal row
11. Atomic mass is the weighted average mass of all the naturally occurring isotopes of the element.  
Atomic number is number of protons found in the nucleus of each atom of the element.
12. Ar & K; Co & Ni; Th & Pa; U & Np; Pu & Am; Sg & Bh

### Section 2:

1. Identify the element & write its noble gas notation electron configuration:
  - a. Germanium; Ge [Ar]  $4s^2 3d^{10} 4p^2$
  - b. Bismuth; Bi [Xe]  $6s^2 4f^{14} 5d^{10} 6p^3$
  - c. Scandium; Sc [Ar]  $4s^2 3d^1$
  - d. Radium; Ra [Rn]  $7s^2$
2. Identifying off the periodic table:
  - a. p-block
  - b. B = Alkali Metals  
C = Alkaline Metals  
D = Transition Metals  
E = Main-Group Elements  
F = Halogens  
G = Noble Gases  
H = Actinides
3. Give the symbol, period, group, & block:
  - a. Sulfur; S, 3<sup>rd</sup> Period, Group 16, p-block
  - b. Nickel; Ni, 4<sup>th</sup> Period, Group 10, d-block
  - c. ? [Kr] 5s1; Rb, 5<sup>th</sup> Period, Group 1, s-block
  - d. ? [Ar] 3d5 4s1; Cr, 4<sup>th</sup> Period, Group 6, d-block      \* Tricky one with an aufbau exception configuration!
4. Group #s of blocks:
  - a. s-block: Group 1 & Group 2
  - b. p-block: Groups 13-18
  - c. d-block: Groups 3-12

# Chemistry Chapter 5 Concept Review Worksheet Answers

**Judd E. Hollander**



## Chemistry Chapter 5 Concept Review Worksheet Answers:

**Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998      **Holt Chemistry** Ralph Thomas Myers, 2004      *Teaching Undergraduate Science* Linda C. Hodges, 2023-07-03 This book is written for all science or engineering faculty who have ever found themselves baffled and frustrated by their undergraduate students lack of engagement and learning The author an experienced scientist faculty member and educational consultant addresses these issues with the knowledge of faculty interests constraints and day to day concerns in mind Drawing from the research on learning she offers faculty new ways to think about the struggles their science students face She then provides a range of evidence based teaching strategies that can make the time faculty spend in the classroom more productive and satisfying Linda Hodges reviews the various learning problems endemic to teaching science explains why they are so common and persistent and presents a digest of key ideas and strategies to address them based on the research she has undertaken into the literature on the cognitive sciences and education Recognizing that faculty have different views about teaching different comfort levels with alternative teaching approaches and are often pressed for time Linda Hodges takes these constraints into account by first offering a framework for thinking purposefully about course design and teaching choices and then providing a range of strategies to address very specific teaching barriers whether it be students motivation engagement in class ability to problem solve their reading comprehension or laboratory research or writing skills Except for the first and last chapters the other chapters in this book stand on their own i e can be read in any order and address a specific challenge students have in learning and doing science Each chapter summarizes the research explaining why students struggle and concludes by offering several teaching options categorized by how easy or difficult they are to implement Some for example can work in a large lecture class without a great expenditure of time others may require more preparation and a more adventurous approach to teaching Each strategy is accompanied by a table categorizing its likely impact how much time it will take in class or out and how difficult it will be to implement Like scientific research teaching works best when faculty start with a goal in mind plan an approach building on the literature use well tested methodologies and analyze results for future trials Linda Hodges message is that with such intentional thought and a bit of effort faculty can succeed in helping many more students gain exciting new skills and abilities whether those students are potential scientists or physicians or entrepreneurs Her book serves as a mini compendium of current research as well as a protocol manual a readily accessible guide to the literature the best practices known to date and a framework for thinking about teaching      **Chemistry 'O' Level** Rex M. Heyworth, 2007      First Year Chemistry Students' Conceptions of Acid/base Chemistry Sally Diane Rupert, 2001      Science Insights , 1999      Chemistry insights 'O' level Rex M. Heyworth, 2007      Your Personalized Guide to Academic Success Grace Charles, 2009-08-10 GRACE CHARLES award winning high school student Yeshiva University valedictorian and medical student at Mount Sinai School of Medicine has years of experience tutoring high school and college students But her

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