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(Class 12)

Give structures of the products you would expect when each of the following alcohol reacts with (a) $\text{HCl}-\text{ZnCl}_2$, (b) HBr and (c) SOCl_2 .

(ii) 2-Methylbutan-2-ol

(a) (i) $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \text{OH} \xrightarrow{\text{HCl, ZnCl}_2} \text{No reaction}$

Primary alcohols do not react appreciably with Lucas' reagent ($\text{HCl} - \text{ZnCl}_2$) at room temperature.

(ii)

$$\begin{array}{ccc} \text{OH} & & \text{Cl} \\ | & & | \\ \text{CH}_3 - \text{CH}_2 - \text{C} - \text{CH}_3 & \xrightarrow{\text{HCl-ZnCl}_2} & \text{CH}_3 - \text{CH}_2 - \text{C} - \text{CH}_3 + \text{H}_2 \\ | & & | \\ \text{CH}_3 & & \text{CH}_3 \end{array}$$

2 - Methylbutan - 2 - ol (3°) 2 - Chloro - 2 - Methylbutane

Tertiary alcohols react immediately with Lucas' reagent.

(b) (i) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH} + \text{HBr} \xrightarrow{-\text{H}_2\text{O}} \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{Br}$
Butan-1-ol 1-Bromobutane

(ii)

$$\begin{array}{c} \text{OH} \\ | \\ \text{CH}_3 - \text{CH}_2 - \text{C} - \text{CH}_3 \\ | \\ \text{CH}_3 \end{array} + \text{HBr} \longrightarrow \begin{array}{c} \text{Br} \\ | \\ \text{CH}_3 - \text{CH}_2 - \text{C} - \text{CH}_3 \\ | \\ \text{CH}_3 \end{array} + \text{H}_2\text{O}$$

2-Methylbutan-2-ol(3°) 2-Bromo-2-Methylbutane

(c) (i) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH} + \text{SOCl}_2 \longrightarrow \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{Cl} + \text{SO}_2 + \text{HCl}$
Butan-1-ol 1-chlorobutane

(ii)

$$\text{CH}_3 - \text{CH}_2 - \underset{\begin{array}{c} | \\ \text{CH}_3 \end{array}}{\overset{\begin{array}{c} \text{OH} \end{array}}{\text{C}}} - \text{CH}_3 + \text{SOCl}_2 \rightarrow \text{CH}_3 - \text{CH}_2 - \underset{\begin{array}{c} | \\ \text{CH}_3 \end{array}}{\overset{\begin{array}{c} \text{Cl} \end{array}}{\text{C}}} - \text{CH}_3 + \text{SO}_2 + \text{HCl}$$

2-Methylbutan-2-ol 2-Chloro-2-methylbutane

Question 111
Predict the major product of acid catalysed dehydration of

(iii) butan-1-ol

[11]



(ii) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH} \xrightarrow{\text{Dehydration / H}^+} \text{CH}_3\text{CH}=\text{CHCH}_3 + \text{H}_2\text{O}$
Butan-1-ol But-2-ene
(Major product)

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Padhraic Smyth



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