

## Chemistry Semester 2 Review

### Quarter Three Performance Expectations

#### Unit 10: The Mole

GOT (T)I	Task I should have mastered
	Relate the mole to common counting units such as dozens and pairs.
	Compare the relative sizes of Avogadro's number and common counting units.
	Identify the representative particles of elemental, ionic, and molecular substances.
	Calculate molar mass for compounds and molecules.
	Use molar mass and Avogadro's number to convert mass to moles and moles to mass for pure substances.
	Convert between moles to grams and moles to atoms for pure substances.

#### Unit 11: Chemical Reactions

GOT (T)I	Task I should have mastered
	Analyze chemical equations for evidence of a chemical reaction (production of a gas, precipitate, water, heat/light, or a new substance).
	Show evidence of the law of conservation of mass by writing a chemical equation in terms of moles, mass, or molecules.
	Represent solids (s), liquids (l), gases (g) and aqueous solutions (aq) in chemical reactions.
	Write word equations, skeleton equations, and balanced chemical equations.
	Classify common reactions as combustion, synthesis, decomposition, single replacement, or double replacement.
	Predict the products of hydrocarbon combustion reactions, write the equation, and balance it.
	Apply the mole concept and conservation of mass to chemical reactions.
	Distinguish and define limiting versus excess reactants in a chemical reaction.

#### Unit 12: Thermochemistry

GOT (T)I	Task I should have mastered
	Model energy changes in chemical reactions and relate them to the laws of conservation of energy and matter.
	Relate chemical potential energy to exothermic and endothermic reactions.
	Analyze calorimetry data to determine changes to the system, surroundings, and universe.
	Analyze a time/temperature graph for phase changes.
	Summarize the concepts of heat of vaporization, heat of fusion, and specific heat capacity.
	Predict the of enthalpy for exothermic and endothermic reactions.
	Cite evidence to explain how collision theory relates to reaction rates.
	Summarize the factors that affect reaction rates.

# Chemistry Semester Two Course Review

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