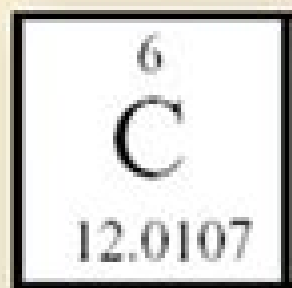


Calculating Formula Mass

Calculate the formula mass of carbon dioxide, CO_2 .



$$12.01 \text{ g} + 2(16.00 \text{ g}) = 44.01 \text{ g}$$

\therefore One mole of CO_2 (6.02×10^{23} molecules) has a mass of 44.01 grams

Chemistry Calculating Formula Mass Answers

**Theodore L. Brown, H. Eugene LeMay
Jr., Bruce E. Bursten, Catherine
Murphy, Patrick Woodward, Steven
Langford, Dalius Sagatys, Adrian
George**

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