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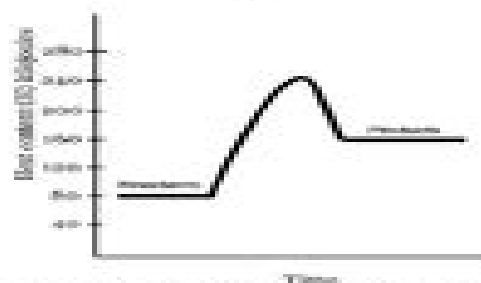
Assessment of Learning (Product)

Multiple Choice; Circle the most correct answer. (15K)

1. Which of the following phase changes involve the removal (release of heat)?
A) condensation B) vaporization C) freezing D) melting E) A & C

2. If the heat of formation for the chemical reaction $\text{Sg(s)} + 12\text{O}_2\text{(g)} \rightarrow 8\text{SO}_3\text{(g)}$, is -395.7 kJ/mol of SO_3 , how can the equation be rewritten to include the heat term?

- A. $\text{Sg(s)} + 12\text{O}_2\text{(g)} \rightarrow 8\text{SO}_3\text{(g)} + 3166 \text{ kJ}$ C. $\text{Sg(s)} + 12\text{O}_2\text{(g)} \rightarrow 8\text{SO}_3\text{(g)} + 395.7 \text{ kJ}$
B. $\text{Sg(s)} + 12\text{O}_2\text{(g)} \rightarrow 8\text{SO}_3\text{(g)} - 395.7 \text{ kJ}$ D. $\text{Sg(s)} + 12\text{O}_2\text{(g)} + 3166 \text{ kJ} \rightarrow 8\text{SO}_3\text{(g)}$



3. In the diagram above, the energy content of the products
A) is the greater than the energy content of the reactants
B) is less than the energy content of the reactants
C) is the same as the energy content of the reactants
D) none of the above

4. The quantity of heat required to change the temperature of 1 g of a substance by 1°C is defined as _____

- A) a joule B) specific heat C) a calorie D) density

5. According to Hess's law of heat summation, which of the following statements is true?

- A) The enthalpy change is independent of the number of intermediate steps.
B) The enthalpy change depends on the pathway of the process.
C) The enthalpy change is independent of the pathway of the process.
D) Both A and C

Chemistry Ch Enthalpy Test

Joseph A. Mascetta, Mark Kernion

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author names.

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