# CHAPTER 14 STUDY GUIDE

# Mixtures and Solutions

#### Section 14.1 Heterogeneous and Homogeneous Mixtures

In your textbook, read about suspensions and colloids.

For each statement below, write true or false.

Salter	A solution is a mixture containing particles that settle out of the mixture if left undisturbed.	
Trust	<ol><li>The most abundant substance in a colloid is the dispersion medium.</li></ol>	
tatue	<ol><li>A colloid can be separated by filtration.</li></ol>	
true	<ol> <li>A solid emulsion consists of a liquid dispersed in a solid.</li> </ol>	
Tue	<ol><li>Whipped cream is an example of a foam.</li></ol>	
Nation	6. In an aerosol, the dispersing medium is a liquid.	
Tue	<ol><li>Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.</li></ol>	
Tue	<ol> <li>Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces.</li> </ol>	
floitus	<ol><li>Stirring an electrolyte into a colloid stabilizes the colloid.</li></ol>	
Tive	10. Colloids demonstrate the Tyndall effect.	

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
11. Less than 1 nm in diameter	157 150 1	~	$\rightarrow$
12. Between I nm and 1000 nm in diameter	7		
13. More than 1000 nm in diameter		<b>&gt;</b>	
14. Settle out if undisturbed			
15. Pass through standard filter paper			
16. Lower vapor pressure			<b>\</b>
17. Scatter light			

# **Chemistry Chapter 14 Mixtures Solutions Study Guide Answers**

Kenneth W. Raymond

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