

## SOLUTIONS

9. A 1% solution of solute 'X' is isotonic with a 6% solution of sucrose (molar mass =  $342 \text{ g mol}^{-1}$ ). The molar mass of solute 'X' is :  
 (a)  $34.2 \text{ g mol}^{-1}$  (b)  $57 \text{ g mol}^{-1}$   
 (c)  $114 \text{ g mol}^{-1}$  (d)  $3.42 \text{ g mol}^{-1}$  **(1)**
8. On mixing 10 mL of acetone with 50 mL of chloroform, the total volume of the solution will be :  
 (a)  $< 60 \text{ mL}$  (b)  $> 60 \text{ mL}$   
 (c)  $= 60 \text{ mL}$  (d)  $= 100 \text{ mL}$  **(1)**
4. The system that forms maximum boiling azeotrope is :  
 (a)  $\text{CS}_2$  and acetone (b) Benzene and toluene  
 (c) Ethanol and acetone (d) Chloroform and acetone **(1)**
5. Phenol dimerises in benzene having van't Hoff factor 0.54. Its degree of association is:  
 (a) 0.54 (b) 0.46  
 (c) 0.92 (d) 0.27
1. Low concentration of oxygen in the blood and tissues of people living at high altitudes is due to :  
 (a) Low temperature (b) Low atmospheric pressure  
 (c) High atmospheric pressure (d) High temperature **(1)**
4. For an electrolyte undergoing association in a solvent, the van't Hoff factor :  
 (a) is always greater than one  
 (b) has negative value  
 (c) has zero value  
 (d) is always less than one **(1)**
8. An azeotropic solution of two liquids has boiling point lower than either of them when it :  
 (a) is saturated  
 (b) shows positive deviation from Raoult's law  
 (c) shows negative deviation from Raoult's law  
 (d) shows no deviation from Raoult's law **(1)**
9. The relative lowering of vapour pressure of an aqueous solution containing non-volatile solute is 0.0225. The mole fraction of the non-volatile solute is :  
 (a) 0.80 (b) 0.725  
 (c) 0.15 (d) 0.0225 **(1)**
- 10\*. A 1% solution of solute 'X' is isotonic with a 6% solution of sucrose (molar mass =  $342 \text{ g mol}^{-1}$ ). The molar mass of solute 'X' is :  
 (a)  $34.2 \text{ g mol}^{-1}$  (b)  $57 \text{ g mol}^{-1}$   
 (c)  $114 \text{ g mol}^{-1}$  (d)  $3.42 \text{ g mol}^{-1}$  **(1)**

Question numbers 13 to 16 are assertion reason types.

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