

# STOICHIOMETRY WORKSHEET

1. Calculate the number of moles of NaOH that are needed to react with 500 g of  $\text{H}_2\text{SO}_4$  according to the following equation:



2. Calculate the mass of  $\text{NH}_3$  that can be produced from the reaction of 125 g of  $\text{NCl}_3$  according to the following equation:



3. Identify the limiting reactant and determine the mass of  $\text{CO}_2$  that can be produced from the reaction of 25 g of  $\text{C}_3\text{H}_8$  with 75 g of  $\text{O}_2$  according to the following equation:



4. How many grams of  $\text{SO}_2$  are produced when 152 g of  $\text{CS}_2$  reacts with 48.0 g of  $\text{O}_2$  according to the following equation:



# Chemistry Chapter 1study Guide For Content Mastery Stoichiometry Answers

**Robert C. Smoot,Smoot,Richard G.  
Smith,Jack Price**



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