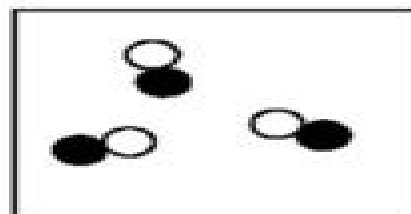
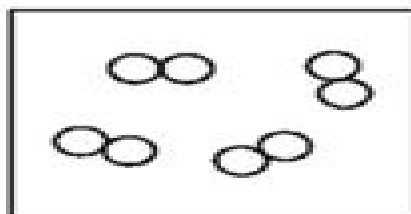
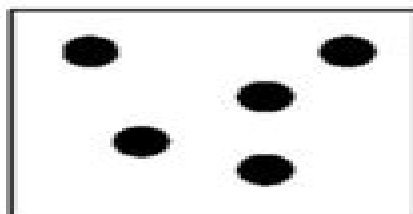
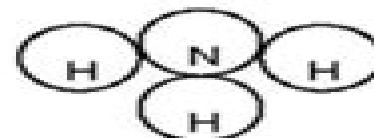
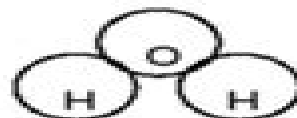
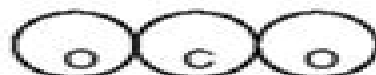
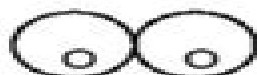


# **Elements, Compounds and Mixtures worksheet**

- Which of the boxes contains:
  - Molecules of an element?
  - Molecules of a compound?
  - Atoms of an element?
  - Do any of the boxes contain a mixture? Explain your answer.



- Look at the following substances. Circle the elements and underline the compounds
  - Li
  - N<sub>2</sub>
  - CO
  - Cl<sub>2</sub>
  - Se
  - NaCl
- How many elements make up each compound?
  - HCl
  - CaCO<sub>3</sub>
  - HNO<sub>3</sub>
  - H<sub>2</sub>SO<sub>4</sub>
  - CH<sub>3</sub>OH
- How many atoms are there in each of these molecules?
  - N<sub>2</sub>
  - CO<sub>2</sub>
  - O<sub>3</sub>
  - CH<sub>4</sub>
  - SO<sub>2</sub>
  - C<sub>2</sub>H<sub>4</sub>
- Circle the molecule with the most atoms, in each pair;
  - HCl or POCl
  - NH<sub>3</sub> or HNO<sub>3</sub>
  - C<sub>3</sub>H<sub>8</sub> or C<sub>2</sub>H<sub>5</sub>OH
  - CH<sub>3</sub>COCH<sub>3</sub> or C<sub>6</sub>H<sub>6</sub>
- What is the formula for each molecules shown below



# Chemistry Elements And Compounds Study Guide Answers

**F. Albert Cotton, Geoffrey  
Wilkinson, Paul L. Gaus**



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