

Chemistry EOC Review Packet

Matter and Changes: Pages 34-50

Classifying:

- a. Chemical change
- b. Chemical property
- c. Physical change
- d. Physical property
- e. Heterogeneous mixture
- f. Homogeneous mixture
- g. Compound
- h. Element
- i. Ion
- ii. Liquid
- iii. Solid
- iv. Amount/Concentration of Mass

Concept Questions:

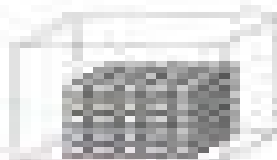
1. Write the definition of matter. Give 3 examples that illustrate what is matter. Include the definition.

Anything that has mass and takes up space. Matter can be described, measured, kept, found.

2. Fill in the chart with whether Matter has it or not. (use the chart that is provided)

Amount of Matter	Volume	Shape
Solid	constant	constant
Liquid	constant	variable
Gas	variable	variable

3. Which picture represents each state of matter?



Solid



Liquid



Gas

4. Give 3 properties of physical properties.

the amount, the mass, the volume, the density, the color, the taste.

5. Give 3 properties of chemical properties.

the ability to burn, react, the ability to combine with other elements.

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Zumdahl

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