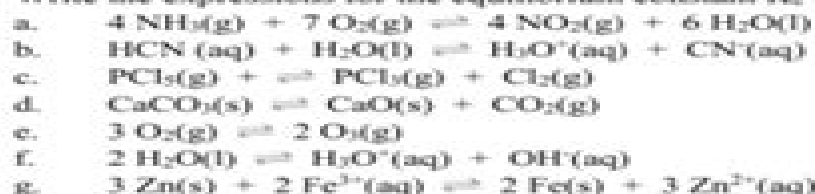


**Chemical Equilibrium****STUDY QUESTIONS & PROBLEMS**

1. Write the expressions for the equilibrium constant
- $K_c$
- for the following reactions:



2. Write the equilibrium constant expressions for the following reactions. How are they related to one another?



3. Calculate the value of the equilibrium constant for the following system, given the data shown:



Concentrations at equilibrium:

$[\text{H}_2]$	=	1.5 mol Liter <sup>-1</sup>
$[\text{CO}_2]$	=	2.5 mol Liter <sup>-1</sup>
$[\text{H}_2\text{O}]$	=	0.5 mol Liter <sup>-1</sup>
$[\text{CO}]$	=	3.0 mol Liter <sup>-1</sup>

4. Chlorine molecules will dissociate at high temperatures into chlorine atoms. At 3000°C, for example,
- $K_c$
- for the equilibrium shown is 0.55. If the partial pressure of chlorine molecules is 1.5 atm, calculate the partial pressure of the chlorine atoms:



5. Suppose that 0.50 moles of hydrogen gas, 0.50 moles of iodine gas, and 0.75 moles of hydrogen iodide gas are introduced into a 2.0 Liter vessel and the system is allowed to reach equilibrium.

Calculate the concentrations of all three substances at equilibrium. At the temperature of the experiment,  $K_c$  equals  $2.0 \times 10^{-2}$ .

6. If the mechanism of a chemical equilibrium consists of two reversible elementary steps, each with its own equilibrium constant
- $K_{c1}$
- and
- $K_{c2}$
- , what expression relates the equilibrium constant
- $K_c$
- for the overall equilibrium to the two constants
- $K_{c1}$
- and
- $K_{c2}$
- ?

7. When 2.0 mol of carbon disulfide and 4.0 mol of chlorine are placed in a 1.0 Liter flask, the following equilibrium system results. At equilibrium, the flask is found to contain 0.30 mol of carbon tetrachloride. What quantities of the other components are present in this equilibrium mixture?



# Chemistry Equilibrium Practice Problems Calculation

**Robert R. Redfield**



## **Chemistry Equilibrium Practice Problems Calculation:**

Chemistry Equilibrium Practice Problems Calculation: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous compelling novels enthralling the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the captivating narratives that have captivated audiences this year. Chemistry Equilibrium Practice Problems Calculation : Colleen Hoover's "It Ends with Us" This heartfelt tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. Chemistry Equilibrium Practice Problems Calculation : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens' "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is an exceptional and thrilling novel that will keep you wondering until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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