#### Gases Test SCH3U

Name:

\_\_\_K \_\_T \_\_C \_\_A Assessment of Learning (Product)
15 9 5 10

#### Multiple Choice. Circle the most correct answer. (15K)

1. Convert the pressure 1.80 kPa to atm.

- A) 2 atm B) 115 atm C) 132 at
  - C) 132 atm D) 245 atm
- E) None of the above

2. What happens to the temperature of a gas when it is expanded quickly?



3. If all of the following flasks are the same size, at the same temperature, and contain the same number of molecules, in which flask will the pressure be highest?

- A) All have the same pressure
- B) Flask I

C) Flask 2

D) Flask 3

E) Flask 4

4. Which diagram best represents a gas in a closed container?









According to the kinetic molecular theory, the Kelvin temperature of an ideal gas is proportional to which one of the following:

A) acceleration

- B) momentum
- C) kinetic energy
- D) potential energy

6. A gas is \_\_\_\_\_ and assumes \_\_\_\_ of its container whereas a liquid is \_\_\_\_ and assumes \_\_\_\_ of its container.

- A) compressible, the volume and shape, not compressible, the shape of a portion
- B) compressible, the shape, not compressible, the volume and shape
- C) compressible, the volume and shape, compressible, the volume
- D) condensed, the volume and shape, condensed, the volume and shape
- E) condensed, the shape, compressible, the volume and shape

# **Chemistry Gases Unit Study Guide Answers**

**L** Cohen

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# **Table of Contents Chemistry Gases Unit Study Guide Answers**

- 1. Understanding the eBook Chemistry Gases Unit Study Guide Answers
  - The Rise of Digital Reading Chemistry Gases Unit Study Guide Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry Gases Unit Study Guide Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - $\circ\,$  Features to Look for in an Chemistry Gases Unit Study Guide Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry Gases Unit Study Guide Answers
  - Personalized Recommendations
  - Chemistry Gases Unit Study Guide Answers User Reviews and Ratings
  - Chemistry Gases Unit Study Guide Answers and Bestseller Lists
- 5. Accessing Chemistry Gases Unit Study Guide Answers Free and Paid eBooks
  - Chemistry Gases Unit Study Guide Answers Public Domain eBooks
  - Chemistry Gases Unit Study Guide Answers eBook Subscription Services
  - Chemistry Gases Unit Study Guide Answers Budget-Friendly Options
- 6. Navigating Chemistry Gases Unit Study Guide Answers eBook Formats

- o ePub, PDF, MOBI, and More
- Chemistry Gases Unit Study Guide Answers Compatibility with Devices
- Chemistry Gases Unit Study Guide Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Chemistry Gases Unit Study Guide Answers
  - Highlighting and Note-Taking Chemistry Gases Unit Study Guide Answers
  - Interactive Elements Chemistry Gases Unit Study Guide Answers
- 8. Staying Engaged with Chemistry Gases Unit Study Guide Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Gases Unit Study Guide Answers
- 9. Balancing eBooks and Physical Books Chemistry Gases Unit Study Guide Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Gases Unit Study Guide Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Gases Unit Study Guide Answers
  - Setting Reading Goals Chemistry Gases Unit Study Guide Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Gases Unit Study Guide Answers
  - Fact-Checking eBook Content of Chemistry Gases Unit Study Guide Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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