

## Predicting Products KEY

**Step 1:** Identify the type of chemical reaction that is occurring.

**Types of Chemical Reactions:**

**SYNTHESIS:** 2 or more elements combine to form a compound



**DECOMPOSITION:** A compound breaks down into 2 or more elements



**SINGLE REPLACEMENT:** One element takes the place of another element in a compound



"Cations (positive ions) switch places"



**DOUBLE REPLACEMENT:** Two elements change places in each of their compounds



**ACID-BASE NEUTRALIZATION:** A type of double replacement involving the reaction of an acid (includes  $H^+$  ions) with a base (includes  $OH^-$  ions) to form a salt (group 2 + group 17) and  $H_2O$



**COMBUSTION:** Decomposition of a hydrocarbon (contains C and H in some quantity and sometimes O) in the presence of oxygen; always produces  $CO$  or  $CO_2$  and  $H_2O$



**Give it a try!** Identify the following types of chemical reactions.



Type of Reaction: **Double Replacement**



Type of Reaction: **Combustion**



Type of Reaction: **Decomposition**



Type of Reaction: **Single Replacement**



Type of Reaction: **Acid-Base Neutralization/Double Replacement**



Type of Reaction: **Synthesis**

# Chemistry Predicting Products Packet Key

**RD Boyd**



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