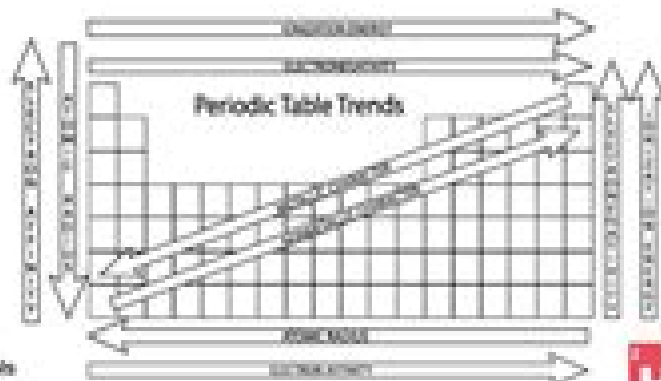


# PERIODIC TABLE OF THE ELEMENTS

Atomic Number → **1** ← Symbol  
 Name → **Hydrogen** ← Atomic Weight

○ Solid  
 ☹ Liquid  
 ☼ Gas  
 \*At room temperature

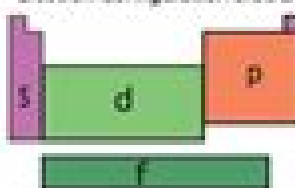


⬇ Magnetic  
 ⬇ Noble metals  
 ⬇ Radioactive  
 ⬇ Only traces found in nature  
 ⬇ Synthetic

Alkali Metals      Non-Metals  
 Alkaline Earth Metals      Halogens  
 Transition Metals      Noble Gases  
 Other Metals      Lanthanides  
 Metalloids      Actinides

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 <b>H</b> Hydrogen 1.008	2 <b>He</b> Helium 4.003																
3 <b>Li</b> Lithium 6.941	4 <b>Be</b> Beryllium 9.012																10 <b>Ne</b> Neon 20.180
5 <b>Na</b> Sodium 22.990	6 <b>Mg</b> Magnesium 24.305																18 <b>Ar</b> Argon 39.948
19 <b>K</b> Potassium 39.098	20 <b>Ca</b> Calcium 40.078	21 <b>Sc</b> Scandium 44.956	22 <b>Ti</b> Titanium 47.88	23 <b>V</b> Vanadium 50.942	24 <b>Cr</b> Chromium 52.00	25 <b>Mn</b> Manganese 54.938	26 <b>Fe</b> Iron 55.845	27 <b>Co</b> Cobalt 58.933	28 <b>Ni</b> Nickel 58.693	29 <b>Cu</b> Copper 63.546	30 <b>Zn</b> Zinc 65.38	31 <b>Ga</b> Gallium 69.723	32 <b>Ge</b> Germanium 72.64	33 <b>As</b> Arsenic 74.922	34 <b>Se</b> Selenium 78.96	35 <b>Br</b> Bromine 79.904	36 <b>Kr</b> Krypton 83.798
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**Cecil Hobart Peabody**



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