

### Section 3: Chemistry

41. The number of electrons in an atom with atomic number 105 having  $(n + l) = 8$  are:  
 (a) 30 (b) 17 (c) 15 (d) 18
42. The ratio  $\frac{a}{b}$  ( $a$  and  $b$  being the van der Waal's constant of real gases) has the dimensions of  
 (a)  $\text{atm mol}^{-1}$  (b)  $\text{L mol}^{-1}$  (c)  $\text{atm L mol}^{-1}$  (d)  $\text{atm L mol}^{-2}$
43. Resultant molarity of  $\text{H}^+$  ion in a mixture of 100 mL of 0.1 M  $\text{H}_2\text{SO}_4$  and 200 mL of 0.1 M  $\text{H}_3\text{PO}_4$  is:  
 (a) 0.1 M (b) 0.2 M (c) 0.267 M (d) 0.133 M
44. The chemical reaction,  $2\text{O}_3 \longrightarrow 3\text{O}_2$  proceeds as follows:  
 $\text{O}_3 \xrightarrow{\text{fast}} \text{O}_2 + \text{O} \quad \dots \text{ (fast)}$   
 $\text{O} + \text{O}_3 \longrightarrow 2\text{O}_2 \quad \dots \text{ (slow)}$   
 The rate law expression should be:  
 (a)  $r = k[\text{O}_3]^2$  (b)  $r = k[\text{O}_3]^2 [\text{O}_2]^{-1}$  (c)  $r = k [\text{O}_3] [\text{O}_2]$  (d) unpredictable
45. For the reaction in equilibrium,  
 $\text{N}_2\text{O}_4(\text{g}) \rightleftharpoons 2\text{NO}_2(\text{g})$   
 the concentrations, of  $\text{N}_2\text{O}_4$  and  $\text{NO}_2$  at equilibrium are  $4.8 \times 10^{-2}$  and  $1.2 \times 10^{-2} \text{ mol L}^{-1}$  respectively. The value of  $K_c$  for this reaction is:  
 (a)  $3 \times 10^{-3} \text{ mol L}^{-1}$  (b)  $3 \times 10^3 \text{ mol L}^{-1}$  (c)  $3.3 \times 10^2 \text{ mol L}^{-1}$  (d)  $3 \times 10^{-1} \text{ mol L}^{-1}$
46. Which one of the following is tribasic acid?  
 (a)  $\text{H}_3\text{PO}_2$  (b)  $\text{H}_3\text{PO}_3$  (c)  $\text{H}_3\text{PO}_4$  (d)  $\text{H}_4\text{P}_2\text{O}_7$
47. In CsBr crystal structure edge length of unit cell is  $4.3 \text{ \AA}$ . The shortest interionic distance between  $\text{Cs}^+$  and  $\text{Br}^-$  ions is  
 (a)  $3.72 \text{ \AA}$  (b)  $1.86 \text{ \AA}$  (c)  $7.44 \text{ \AA}$  (d)  $4.3 \text{ \AA}$

# Chemistry Section 3 Mastering Problems Answers

**John S. Phillips, Cheryl Wistrom**



### **Chemistry Section 3 Mastering Problems Answers:**

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