

Name _____

Alternate Assignment to Collage Creation

Chemistry Webquest #1: Introduction to Atoms and Atomic Scientists

You must complete all 3 Parts!!

Directions - Part I – A BRIEF History of the Atom

Click the following link: <http://www.timequest.com/teachers/atomic-scientist/>

Use the information in this web page to fill in your History of the Atom Timeline. There are many additional activities that contributed to the atomic model. I only expect you to know about six of them.

History of the Atom Timeline

	1	2	3	4	5	6
Date:	400 BC		1897	1911		1930
Scientist:		John Dalton			Niels Bohr	

Directions - Part II – Models of the Atom

Click the following link: <http://www.universetoday.com/69058/democritus-atom/>

1. Where was Democritus from? _____
2. Why is he important with regard to atomic theory? _____

Click the following link: <http://www.cis.edu/~cpaulbd/C311webpages/composition/dalton.html>

1. What are the four parts to Dalton's Theory?
•
•
•
•
2. Why is he important with regard to atomic theory? _____

Click the following link: <https://education-portal.com/academy/lesson/early-atomic-theory-dalton-atoms-post-riohardest-math-and-likemath.com/>

1. What did JJ Thomson discover? _____
2. What was the name of the equipment (experiment) he did? _____
3. What "name" is given his model? _____ . Describe his model in the space below.

Chemistry Solutions Webquest Answer Key

Matthew J. Johll



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Questions and Solutions for Chemistry John Sadler, Mike Taylor, 1986 *Complete Solutions and Answers for General Chemistry* John Mays, 2018-01-29 All answers to exercises quizzes and tests plus worked out solutions General Chemistry uses a mastery learning method In this method students build comprehension by adding new concepts while reviewing and rehearsing key material throughout the year This method is implemented in carefully crafted exercises quizzes and the textbook narrative and it facilitates learning mastery and retention Students appreciate the smaller profile and lighter weight of our books This is possible because of aspects of our Textbook Philosophy When mastery is the goal you must cull down the material to an amount that can reasonably be mastered in one year Related subjects are integrated into the narrative The history of modern chemistry mathematics and technical communication is emphasized throughout Integration protects a course from seeming compartmentalized from other subjects That s certainly not how the real world is Real chemists use math and writing skills every day Their knowledge about how the scientific enterprise works is greatly enhanced by their knowledge of great scientists who came before Our approach to implementing a Kingdom Perspective does not entail unnaturally inserting Bible verses or devotional insets A better approach for Christians is to do science from the faith informed perspective of disciples of Christ studying God s created order This is apparent in General Chemistry from the observations about order in the universe in how nature submits to scientific study and modeling and in the fitness of creation as a habitat for humanity and animal life **Chemistry** Steven S. Zumdahl, 1989 *Essential Chemistry* Brandon J. Cruickshank, 2000 **Student Solution Manual for Introduction to Chemical Principles** H. Stephen Stoker, Nancy J. Gardner, 2013-04-22 The Student Solutions Manual includes full solutions to all odd numbered end of chapter problems in the text and answers to all multiple choice practice test questions **Chemistry, Student Solutions Manual** John A. Olmsted, Gregory M. Williams, 2002-03-22 This third edition continues to innovate by providing students with an integrated and modern approach to the subject The text emphasizes the modern tools of chemistry while incorporating historical evidence and its unique molecular quantitative emphasis is further reinforced by an integrated media package developed by the authors Also of benefit is the just in time presentation of key content only providing details once they are needed While key topics and analytical techniques have been updated there is now an additional third chapter on chemical equilibrium The authors have also developed an expanded and more integrated problem solving emphasis that now incorporates a 4 step strategy throughout complete with text icons The whole is backed by a range of supplements including a new illustration program a tutorial CD interactive learningware an extensive Web CT component an instructor s resource CD and a solution CD **Selected Solutions Manual [for] Principles of Chemistry, a Molecular Approach, Third Edition [by] Nivaldo J. Tro** Nivaldo J. Tro, Kathy Thrush Shaginaw, 2015-01-13 The selected solution manual for students contains complete step by step solutions to selected odd numbered end of chapter problems *The Answer Key: A Comprehensive Explanation of*

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Solutions Guide for Chemistry Steven S. Zumdahl, Kenneth C. Brooks, 1989 Chemistry with Student Solution Guide 6th Edition Zumdahl, 2003-01-01

Introductory Chemistry Selected Solutions Manual Matthew J. Johll, 2008 This provides solutions only to those problems that have a short answer in the text s Answers section problems numbered in blue in the text

Chemistry Solutions, 2009

Study Guide and Solutions Manual to accompany Basic Concepts of Chemistry, 9th Edition Leo J. Malone, 2011-12-07 This is the Study Guide and Solutions Manual to accompany Malone s Basic Concepts of Chemistry

Chemistry Test Key Units 1-10 (RES) Responsive Education Solutions Staff, 2011-08-01 Key Individual Test Key for Chemistry Units 1 10

Selected Solutions Manual for Principles of Chemistry Nivaldo Tro, Kathleen Shaginaw, 2019-03 The selected solution manual for students contains complete step by step solutions to selected odd numbered end of chapter problems

Student Solutions Manual for Moore/Stanitski's Chemistry: the Molecular Science, 5th Judy Ozment, John W. Moore, Conrad L. Stanitski, 2014-04-07 Go beyond the answers to truly understanding the steps it takes to get there This solutions manual contains fully worked out solutions to end of chapter questions that have blue boldface numbers and are answered in the back of the text Solutions match the problem solving strategies used in the main text

Chemistry Phillip John Buddle, J. S. Mackay, 1970

Solutions Manual for Chemistry John McMurry, Robert Fay, Joseph Topich, Jill Robinson, 2015-06-19 Contains solutions to all in chapter problems and solutions to even numbered end of chapter problems

General Chemistry - Solutions Manual P. W. Atkins, Edward Atkins, 1989-05

Chemistry Steven S. Zumdahl, 1996-12

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