Use with Section 5.3





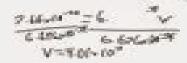




Figure 1 The emission spectrum and orbit transition diagram for hydrogen.

#### Tabble II.

Transition	E (2)	w (s:")	2(m)	Type of Radispins
a~6-a-5	2.66 × 10 <sup>-06</sup>	4.0126	7.48×10-6	Trickey
a = 6 a = 4	2.57 = 10 <sup>-20</sup>	1.14 × 10 "	2.63x16"	Irtueh
m=6-m=3	1.82 × 10 ***	2.15×10"	1.01210.	Intract
$\alpha=6 \cdots \alpha=2$	4.84 × 10 ***	7.30 x10	d-11710.	Infred
$\alpha = 0 \longrightarrow \alpha = 1$	2.12 × 80 °F	3.70×10'>	937×10	visible
n=5 - m + 4	$4.93 \times 10^{-10}$	7.4MB	4.01210-	Infred.
a=5-a=3	1.55 × 10°	2-34XIC	1.28,10	Infind
$\alpha=3\cdots\alpha=2$	4.56 × 10 <sup>(1)</sup>	6.84×11,	4.36×10-	Ihand
$\alpha=5 \longrightarrow \alpha=1$	2.09 × 80 <sup>cm</sup>	3.15×105	9.52,10	Visable
	1.06 × 10°	7-10×1014	1.87.10-6	Inloud
	4.09 × 10 ***	6.17×10N	4.86×10-	Lafat
$m=A \longrightarrow m=1$	2.04 × 10 <sup>-10</sup>	3.08 > 10.2	9.70 10.8	Visibe
$n=3 \leadsto n=2$	$3.00 \times 10^{-10}$	4.58,10"	6.53,107	Infact
$n=3\cdots n-1$	1.544 = 30 -10	2.43×16"	1.02×10-7	Visible
n = 2 n = 1	1.64 × 10°	2.48×10'	1.20×16	Visible





# **Chemistry Interpreting Graphics Answers Key**

Jianjun Gao

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