

Chapter 2: The Chemistry of Life

Learning Standards

Central Concept: Chemical elements form organic molecules that interact to perform the basic functions of life.

- 1.1 Recognize that biological organisms are composed primarily of many few elements. The six most common are Carbon, Hydrogen, Nitrogen, Oxygen, Phosphorus, and Sulfur.
- 1.2 Describe the basic molecular structures and primary functions of the four major categories of organic molecules (carbohydrates, lipids, proteins, and nucleic acids).
- 1.3 Explain the role of enzymes as catalysts that lower the activation energy of biochemical reactions. Identify factors, such as pH and temperature, that have an effect on enzymes.

Key Terms

Atom	Adhesion	Monosaccharide
Nucleus	Mixture	Lipid
Proton, Neutron, Electron	Solution	Nucleic Acid
Element	Solute	Nucleotide
Isotope	Solvent	Protein
Compound	Suspension	Amino Acid
Ionic Bond	pH Scale	Chemical Reaction
Ion	Acid	Reactant
Covalent Bond	Base	Product
Molecule	Buffer	Activation Energy
Van der Waals Forces	Monomer	Catalyst
Hydrogen Bond	Polymer	Enzyme
Cohesion	Carbohydrate	Substrate

You should be able to:

1. Identify the three subatomic particles found in atoms.
2. Explain how all of the isotopes of an element are similar and how they are different.
3. Explain how compounds are different from their component elements.
4. Describe the two main types of chemical bonds.
5. Discuss the unique properties of water.
6. Differentiate between solutions and suspensions.
7. Explain what acidic solutions and basic solutions are.
8. Describe the unique qualities of carbon.
9. Describe the structures and functions of each of the four groups of macromolecules.
10. Explain how chemical reactions affect chemical bonds.
11. Describe how energy changes affect how easily a chemical reaction will occur.
12. Explain why enzymes are important to living things.

Chemistry Of Life Study Guide Answers

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Life Study Guide David E. Sadava, Gordon H. Orians, Craig Heller, William K. Purves, 2006-12-22 Especially helpful for AP Biology students each chapter of the study guide offers a variety of study and review tools The contents of each chapter are broken down into both a detailed review of the Important Concepts covered and a boiled down Big Picture snapshot The guide also covers study strategies common problem areas and provides a set of study questions both multiple choice and short answer

Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition Susan McMurry, 2000 Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three chapters more detailed explanations in solutions and chapter outlines

Study Guide and Full Solutions Manual Susan McMurry, John McMurry, 2002-08 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all end of chapter problems in the textbook

Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter

Study Guide with Selected Solutions Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

Chemistry and Life John William Hill, Dorothy M. Feigl, Stuart J. Baum, 1993 **Study Guide and Partial Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry** Susan McMurry, 1996 This internationally acclaimed detective series is just the thing for lovers of those Number One Ladies looking for a darker more realistic view of Botswana Sue Baker Publishing News **Study Guide for Life on Earth** Carl S. Pike, 1973 *Study Guide for Atoms, Molecules, and Life* Michael S. Matta, 1981 *Resources in Education*, 1984 **Student Study Guide for Biology [by]**

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Student Study Guide to Accompany Human Biology Sylvia S. Mader, 2003-08 This best selling text emphasizes the relationship between humans and other living things Intended for an introductory course this text provides students with a firm grasp of how their bodies function and how the human population can become more fully integrated into the biosphere An Online Learning Center tied directly to the text via icons will direct students to activities or animations that gives a visual example of difficult processes as well as Working Together boxes to emphasize homeostasis

The Publishers' Trade List Annual, 1893 *Study Guide/Selected Solutions Manual* Julie R.

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