

Test Bank for Chemistry: Structure and Properties, 2nd Edition, Nivaldo J. Tro

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Chemistry: Structure & Properties, 2e (Tro) Chapter 1: Atoms

1.1 Multiple Choice Questions

1) An element has three stable isotopes with masses of 27.977 amu, 28.976 amu, and 29.973 amu. The heaviest two isotopes have an abundance of 4.68% and 3.09%, respectively. What is the mass of the element?

- A) 27.684 amu
- B) 29.251 amu
- C) 28.985 amu
- D) 30.107 amu
- E) 28.991 amu

Answer: C

Diff: 3 Var: 1 Page Ref: 1.9

LO: 1.6

Global: G4

2) All of the following will lead to a homogeneous mixture EXCEPT

- A) sugar and coffee.
- B) mud and water.
- C) salt and water.
- D) ice tea and lemonade.
- E) vinegar and water.

Answer: B

Diff: 1 Var: 1 Page Ref: 1.2

LO: 1.1

Global: G2

3) How are gases different from solids and liquids?

- A) Gases can only be made up of atoms.
- B) The particles in a gas attract each other much more strongly than in solids and liquids.
- C) Gases are compressible.
- D) Only gases can take the shape of their container.
- E) Gases are colorless.

Answer: C

Diff: 1 Var: 1 Page Ref: 1.2

Chemistry Questions And Answers By Tro

Xiaolong Qi



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