

## Ch. 9 Practice Test

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### Chapter: Stoichiometry

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. A balanced chemical equation allows one to determine the
- mole ratio of any two substances in the reaction.
  - energy released in the reaction.
  - electron configuration of all elements in the reaction.
  - mechanism involved in the reaction.
- \_\_\_\_\_ 2. The coefficients in a chemical equation represent the
- masses, in grams, of all reactants and products.
  - relative numbers of moles of reactants and products.
  - number of atoms in each compound in a reaction.
  - number of valence electrons involved in the reaction.
- \_\_\_\_\_ 3. To determine the limiting reactant in a chemical reaction involving known masses of the two reactants, which of the following would be most useful?
- determining the masses of 100 mol of each reactant
  - determining the molar masses of the products
  - calculating bond energies
  - calculating the mass of a single product formed from each reactant
- \_\_\_\_\_ 4. How many mole ratios can be correctly obtained from the chemical equation  $2\text{Al}_2\text{O}_3(l) \rightarrow 4\text{Al}(s) + 3\text{O}_2(g)$
- 3
  - 4
  - 6
  - 8

# Chemistry Chapter 9 Test Answers

**Claire Aldridge, Karla Lee, Kaplan, Inc**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the text bar.

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