

## ANSWER KEY

### Chemistry Semester 3 Study Guide

This is a cumulative test for tested topics through all three semesters of high school chemistry. Some test topics are covered in previous and subsequent semesters. Some topics are covered previously but not previously tested.

#### Unit 1: Bonding

1. Complete the following table for properties of ionic, covalent, and metallic bonds.

Property	Ionic Bonds	Covalent Bonds	Metallic Bonds
Appearance	crystalline	opaque solid liquid or gas at room temperature	shiny metallic luster
Melting Point (high low, variable)	high	low	medium
Solubility in water	soluble	most are insoluble	most insoluble
Conductivity (as a solid)	no	no	yes
Conductivity (as a liquid)	yes	no	yes
Electrolysis (decomposes in water)	yes	no	yes
Electron location in atoms, transferred, sea?	transferred	electron pairs	in sea
Structure in solid, molecular, crystal lattice, network?	crystal	network	network (close packed)
Electrolysis (melting breaks in water)	most ionic compounds forming neutral molecules	no	electrolysis
Other characteristics (magnet)	most	no	yes
Draw the structure (one molecule)			

2. Complete the following table for properties of ionic, covalent, and metallic bonds.

# Chemistry Cp Study Guide Spring Semester Answers

**Daniel F McAuley**



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