- Caustic soda is obtained on industrial scale by Hg cell which involves the electrolysis of concentrated aqueous solution of sodium chloride using _____ anode and _____ cathode.
 - a. Ti and Hg
 - b. Hq and Ti
 - c. Hg
 - d. Cu, Ti
- 2. A standard hydrogen electrode has zero electrode potential because:
 - a. It is easy to oxidized hydrogen
 - b. The electrode potential is arbitrarily taken as zero which serves as standard.
 - c. Hydrogen atom has only one electron
 - d. Hydrogen is the lightest element
- The cell in which copper electrode which is dipped in 1M CuSO₄ is connected with SHE through external circuit has the cell potential:
 - a. OV
 - b. +1.10 V
 - c. +0.34V
 - d. -0.34V
- 4. Which of the following element has highest positive standard reduction potential (E°) at 298K?
 - a. Li
 - b. Cu
 - c. Au
 - d. Al
- 5. The element which can displace other halogens from their solutions:
 - a. CI
 - b. F
 - c. Br
 - d. I
- 6. The strongest reducing agent of the alkali metal in solution is:
 - a. Li
 - b. Na
 - c. K
 - d. Cs

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