

CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

*Candidates are allowed additional 15 minutes for only reading the paper.
They must NOT start writing during this time.*

This paper is divided into four sections – A, B, C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

Internal choices have been provided in one question each in Section B,

Section C and Section D.

All working, including rough work, should be done on the same sheet as, and adjacent to the rest of the answer.

The intended marks for questions or parts of questions are given in brackets [].

Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

Gas constant $R = 1.987 \text{ cal deg}^{-1} \text{ mol}^{-1} = 8.314 \text{ JK}^{-1} \text{ mol}^{-1}$

$= 0.0821 \text{ dm}^3 \text{ atm K}^{-1} \text{ mol}^{-1}$

$1 \text{ l atm} = 1 \text{ dm}^3 \text{ atm} = 101.3 \text{ J}$ $1 \text{ Faraday} = 96500 \text{ coulombs}$.

Avogadro's number $= 6.023 \times 10^{23}$.

SECTION A – 14 MARKS

Question 1

- (A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1]
brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid, high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into _____ and then into _____.
- (ii) The intermediate activated complex in a chemical reaction is highly _____ due to _____ energy.
- (iii) The coordination number and oxidation state of the complex $\text{K}_4[\text{Fe}(\text{CN})_6]$ are _____ and _____ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives _____ and the reaction is known as _____ reduction.

This Paper consists of 8 printed pages.

Turn over

Chemistry Question Paper For Class 1stateboard

**Donald J. Wink, Sharon Fetzer-
Gislason, Sheila McNicholas**



Chemistry Question Paper For Class 1stateboard:

State Board Questions and Answers Rudolph Max Goepp,1929 Circular of Information United States. Office of Education,1883 **The Neighborhood Youth Corps** Beatrice G. Reubens,United States. Department of Labor. Manpower Administration,1969 **Journal of the American Medical Association** American Medical Association,1917 *Drug Bulletin* ,1928 Report of Hearings of January 17, 1907, on S. 5221, to Regulate the Practice of Osteopathy, to License Osteopathic Physicians, and to Punish Persons Violating the Provisions Therof in the District of Columbia ... United States. Congress. House. Committee on the District of Columbia,1907 **Chemistry** Phillip John Buddle,J. S. Mackay,1970 *Test Your Chemistry* T. I. Morrison,1969 Ordinary Grade Questions and Worked Examples in Chemistry A. W. Mackaill,1971

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Chemistry Part 1 Priyamvada Upadhyay,2023-06-29 CHEMISTRY PART 1 A COMPLETE QUESTION BANK BOOK IS HERE FROM ALL EXAM CRACKERS PUBLICATION IT WILL COVER ALL THE CHAPTERS AND ALL THE IMPORTANT FREQUENTLY ASKED QUESTIONS IT WILL COVER ALL TYPE OF NUMERICALS SHORT ANSWER QUESTIONS LONG ANSWER QUESTIONS AND MOST IMPORTANT QUESTIONS FOR SCHOOL EXAMS THIS TIME WE HAVE INCLUDED A SPECIAL SECTION THAT IS COMPETITIVE CORNER WHERE YOU WILL GET PREVIOUS YEAR QUESTIONS OF JEE AND NEET SEPARATELY *Chemistry* Anthony Tibbertsma,1994 Studying chemistry Questions and answers___ Questions in Chemistry for O and H Grade C. A. Wood,1981-01-01

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Table of Contents Chemistry Question Paper For Class 1stateboard

1. Understanding the eBook Chemistry Question Paper For Class 1stateboard
 - The Rise of Digital Reading Chemistry Question Paper For Class 1stateboard
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Question Paper For Class 1stateboard
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Question Paper For Class 1stateboard
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Question Paper For Class 1stateboard
 - Personalized Recommendations
 - Chemistry Question Paper For Class 1stateboard User Reviews and Ratings
 - Chemistry Question Paper For Class 1stateboard and Bestseller Lists

5. Accessing Chemistry Question Paper For Class 1stateboard Free and Paid eBooks
 - Chemistry Question Paper For Class 1stateboard Public Domain eBooks
 - Chemistry Question Paper For Class 1stateboard eBook Subscription Services
 - Chemistry Question Paper For Class 1stateboard Budget-Friendly Options
6. Navigating Chemistry Question Paper For Class 1stateboard eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Question Paper For Class 1stateboard Compatibility with Devices
 - Chemistry Question Paper For Class 1stateboard Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Question Paper For Class 1stateboard
 - Highlighting and Note-Taking Chemistry Question Paper For Class 1stateboard
 - Interactive Elements Chemistry Question Paper For Class 1stateboard
8. Staying Engaged with Chemistry Question Paper For Class 1stateboard
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Question Paper For Class 1stateboard
9. Balancing eBooks and Physical Books Chemistry Question Paper For Class 1stateboard
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Question Paper For Class 1stateboard
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Question Paper For Class 1stateboard
 - Setting Reading Goals Chemistry Question Paper For Class 1stateboard
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Question Paper For Class 1stateboard
 - Fact-Checking eBook Content of Chemistry Question Paper For Class 1stateboard
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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