

REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

| | |
|----|-----------------|
| H | <i>hydrogen</i> |
| S | <i>sulphur</i> |
| Cl | <i>chlorine</i> |
| C | <i>carbon</i> |
| N | <i>nitrogen</i> |
| Na | <i>sodium</i> |
| Pb | <i>lead</i> |
| Ag | <i>silver</i> |

2. Write the symbols for the following elements.

| | |
|------------|-----------|
| sodium | <i>Na</i> |
| phosphorus | <i>P</i> |
| fluorine | <i>F</i> |
| magnesium | <i>Mg</i> |
| potassium | <i>K</i> |
| calcium | <i>Ca</i> |
| zinc | <i>Zn</i> |
| iron | <i>Fe</i> |

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79** copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5** calcium (Ca) **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

| Element | Atomic Number | Atomic Mass | Protons | Neutrons | Electrons |
|--------------|---------------|-------------|-----------|-----------|-----------|
| Lead | 82 | 207 | 82 | 125 | 82 |
| Argon | 18 | 56 | 18 | 40 | 18 |
| Barium | 56 | 137 | 56 | 61 | 56 |
| Tin | 50 | 112 | 50 | 62 | 50 |

6. Match the unit with its description:

| | | | |
|-----------------|------------------------|----|--------|
| <u>F</u> | 1. amount of substance | A. | joule |
| <u>C</u> | 2. density | B. | g |
| <u>B</u> | 3. mass | C. | g/mL |
| <u>D</u> | 4. molar mass | D. | g/mole |
| <u>A</u> | 5. energy | E. | L |
| <u>E</u> | 6. volume | F. | mole |

Chemistry Section Quiz Answers

T. W. Graham Solomons



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