

Naming Ionic Compounds Practice Worksheet - Solutions

Name the following ionic compounds:

- | | | |
|-----|------------------------------|--------------------------------|
| 1) | NH_4Cl | ammonium chloride |
| 2) | $\text{Fe}(\text{NO}_3)_3$ | iron (III) nitrate |
| 3) | TiBr_3 | titanium (III) bromide |
| 4) | Cu_3P | copper (I) phosphide |
| 5) | SnSe_2 | tin (IV) selenide |
| 6) | GaAs | gallium arsenide |
| 7) | $\text{Pb}(\text{SO}_4)_2$ | lead (IV) sulfate |
| 8) | $\text{Be}(\text{HCO}_3)_2$ | beryllium bicarbonate |
| 9) | $\text{Mn}_2(\text{SO}_3)_3$ | manganese (III) sulfite |
| 10) | $\text{Al}(\text{CN})_3$ | aluminum cyanide |

Write the formulas for the following compounds:

- | | | |
|-----|--------------------------|---|
| 11) | chromium (VI) phosphate | $\text{Cr}(\text{PO}_4)_2$ |
| 12) | vanadium (IV) carbonate | $\text{V}(\text{CO}_3)_2$ |
| 13) | tin (II) nitrite | $\text{Sn}(\text{NO}_2)_2$ |
| 14) | cobalt (III) oxide | Co_2O_3 |
| 15) | titanium (II) acetate | $\text{Ti}(\text{C}_2\text{H}_3\text{O}_2)_2$ |
| 16) | vanadium (V) sulfide | V_2S_5 |
| 17) | chromium (III) hydroxide | $\text{Cr}(\text{OH})_3$ |
| 18) | lithium iodide | LiI |
| 19) | lead (II) nitride | Pb_3N_2 |
| 20) | silver bromide | AgBr |

Chemistry Section Review Answers Naming Ions Answers

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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order **Chemistry** John S. Phillips, Cheryl Wistrom, 2000 Organic Chemistry, Student Solution Manual and Study Guide David R. Klein, 2021-03-16 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Solution Manual and Study Guide 4th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry **Foundations of College Chemistry** Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity

Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section

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