
Chemistry (Semester 2) Final Exam Questions and Answers

a statement that uses chemical formulas to show the identities and relative amounts of the substances of a chemical reaction - correct answer-chemical equation

a starting substance in a chemical reaction - correct answer-reactant

as the temperature of a gas increases, the pressure also increases if the volume remains constant - correct answer-gay lussacs law

liquids are less fluid than gases because - correct answer-their intramolecular forces are stronger

identify the type of chemical reaction shown here- $2\text{C}_2\text{H}_6(\text{g}) + 7\text{O}_2(\text{g}) \rightarrow 4\text{CO}_2(\text{g}) + 6\text{H}_2\text{O}(\text{l})$ - correct answer-combustion

identify the type of chemical reaction shown here- $\text{CO}_2(\text{g}) + \text{CaO}(\text{s}) \rightarrow \text{CaCO}_3(\text{s})$ - correct answer-synthesis

at standard temperature and pressure, one mole of any gas occupies _____ of space - correct answer-22.4L

identify the type of chemical reaction shown here, $2\text{H}_2\text{O}(\text{l}) \rightarrow 2\text{H}_2(\text{g}) + \text{O}_2(\text{g})$ - correct answer-decomposition

when the volume of a gas is increased, what happens to the pressure within the gas if the volume is constant - correct answer-pressure decreases

one atmosphere of pressure is equivalent to - correct answer-101.3kPa

identify the type of chemical reaction shown here- $\text{Mg}(\text{s}) + 2\text{AgNO}_3(\text{aq}) \rightarrow 2\text{Ag}(\text{s}) + \text{Mg}(\text{NO}_3)_2(\text{aq})$ - correct answer-single replacement

a reaction in which a compound breaks down into two or more elements of new compounds - correct answer-decomposition reaction

STP is defined as - correct answer-0K, 101.3kPa

a solid produced during a chemical reaction in a solution - correct answer-precipitate

Chemistry Semester 2 Final Exam Answers

Thomas Griffiths



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