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Chemistry Unit 8 - Worksheet 5 Samples of Every Kind of Problem

On a squares short of paper, write a complete solution to each of the problems below. Pollow the procedures outlined in class, including BCA Tables. Be sure to bus your final answer(s). Final answers should have chemical formulas and not observed names.

- Calculate the number of moles of solid potassium chlorate that must decompose to produce solid potassium chloride and 1.8 moles of oxygen gas.
- In a single displacement receion, magnesium metal reacts with hydrochloric acid solution (HCI) to produce magnesium objection and hydrogen gas. How many moles of hydrochloric acid are specied to completely react with 2.43 g of magnesium?
- Ethans, C₂H₆ reacts with oxygen gas to produce curton directed gas and water vapor. What mass of prospen gas is required to react with 2.20 region of others?
- 4. In the copper adiver nitrate reaction, copper metal and aliver nitrate solution react to produce aliver metal and copper(II) nitrate in solution. A student placed a copper wire with a mass of 2.93 g in the reaction test tube. The aliver sitrate solution contained 1.41 g of silver nitrate. She obtained 0.870 g of silver metal. Calculate the percent yield of silver.
- Phosphorus and liquid brumine react vigorously together to form phosphorus tribromide. If 5.0 g of phosphorus and 35 g of bromine react, how many grams of PBr₃ could be produced? Which reactest is the limiting reactant?
- 6. Zine suffide and coygen gas react to form sine oxide and suffar dioxide gas. Determine the amount of sine oxide that should be produced in a reaction between 46.5 g of sine sulfide and 13.3 g of oxygen. What is the mass of the excess reactant?

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Chemistry Size Of Things Worksheet Answers

Sunny Baker, Kim Baker

Chemistry Size Of Things Worksheet Answers:

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