# Calculating Percentage Composition

Calculate the percentage composition of magnesium carbonate, MgCO<sub>3</sub>.

Formula mass of magnesium carbonate: 24.31 g + 12.01 g + 3(16.00 g) = 84.32 g

$$Mg = \left(\frac{24.31}{84.32}\right) \cdot 100 = 28.83\%$$

$$C = \left(\frac{12.01}{84.32}\right) \cdot 100 = 14.24\%$$

$$O = \left(\frac{48.00}{84.32}\right) \cdot 100 = \underline{56.93\%}$$

# <u>Chemistry Solutions Manual Calculate The Percent</u> <u>Composition</u>

**Julius William Sturmer** 

# **Chemistry Solutions Manual Calculate The Percent Composition:**

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