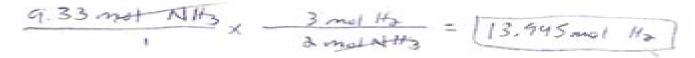
Stoichiometry worksheet #2

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 The equation for the synthesis of ammonia is below. How many moles of H₂ are needed to produce 9.33 moles of NH₃?

N₂ + 3 H₂ → 2.NH₃



 What mass (in grams) of copper is produced when 3.80 g of iron reacts with excess CuSO₄?

Fe + CuSO₄ → FeSO₄ + Cu

If 5.00g of CH₄ are burned according to the equation below, what mass in grams of CO₂ are produced?

CH4 + 2·O2 → CO2 + 2H2O

 In the reaction that follows, determine how many grams of aluminum chloride (AICl₃) will be produced from 92.01 g of Cl₂.

2 AlBr3 + 3 Cl2 → 3 Br2 + 2 AlCl3

Chemistry Stoichiometry Problem Sheet 2 Answers

Ian S. Butler, Arthur E. Grosser

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