

Average Atomic Mass Calculation Worksheet

Answer the following questions.

1. The metal titanium commonly has five isotopes. These are those isotopes, along with the

^{46}Ti - 8%

^{47}Ti - 7.8%

^{48}Ti - 73.4%

^{49}Ti - 5.5%

^{50}Ti - 5.3%

Using this information, determine the average atomic mass of titanium.

2. Complete the table

Isotope	Mass (amu)	Relative Abundance (%)
^{20}Ne	19.992	90.51
^{21}Ne	20.994	
^{22}Ne		9.22

3. Rubidium has two common isotopes, ^{85}Rb and ^{87}Rb . If the abundance of ^{85}Rb is 72.2% and the abundance of ^{87}Rb is 27.8%, what is the average atomic mass of rubidium?

4. Why is the average atomic mass of a ^{12}C atom reported as 12.011 amu in the periodic table?

Average Atomic Mass Worksheet Answers

**Atomic Energy of Canada
Limited, Alastair Graham Walter
Cameron**



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Table of Atomic Masses Nicholas Metropolis, George W. Reitwiesner, 1951 Revised Semiempirical Atomic Mass Formula Atomic Energy of Canada Limited, Alastair Graham Walter Cameron, 1959 Atomic Masses and Fundamental Constants 5 J. Sanders, Aaldert Hendrik Wapstra, 1976-03 **Atomic Masses and Fundamental Constants** 6 Jerry A. Nolen, 2012-12-06 The Sixth International Conference on Atomic Masses was held in East Lansing Michigan Sept 18 21 1979 The conference was initiated organized and sponsored by the Commission on Atomic Masses and Fundamental Constants of the International Union of Pure and Applied Physics The members of the conference committee are listed below W Benenson Chairman Michigan State University R C Barber University of Manitoba E R Cohen Rockwell International Institute of Chemical Physics V I Goldanskii Moscow J C Hardy Chalk River Canada W H Johnson University of Minnesota E Kashy Michigan State University Orsay France R Klapisch J A Nolen Jr Michigan State University R G H Robertson Michigan State University E Roeckl G S I Darmstadt B N Taylor National Bureau of Standards O Schult IKF Julich A H Wapstra IFO Amsterdam N Zeldes Racah Institute Jerusalem The conference was a little different from the preceding one in Paris 1975 in that the fundamental constant aspects were limited to those directly related to atomic masses The gap is to be filled by the second International Conference on Precision Measurement and Fundamental Constants which is now scheduled for June 1981 in Gaithersburg Maryland Only one of the seven sessions in this conference was devoted to fundamental constant determinations The conference was very strongly supported by the Department of Energy the National Science Foundation and the International Union of Pure and Applied Physics **Atomic Masses and Fundamental Constants** 4 J. Sanders, Aaldert Hendrik Wapstra, 1972 The Conference on Nuclear Masses and their Determination which was held at the Max Planck Institute in Mainz from 10 to 12 July 1956 resulted in the formation by the International Union of Pure and Applied Physics of a Commission on Atomic Masses and Related Constants Under the auspices of this Commission conferences covering these subjects were held in Hamilton Ontario 12 16 September 1960 in Vienna 15 19 July 1963 and in Winnipeg 28 August 1 September 1967 After the last of these conferences the Commission felt that the position regarding nuclear masses was reasonably good and that as a consequence the fundamental constants should get more emphasis in future conferences For this reason they were very pleased to accept an offer from the National Physical Laboratory to accept the main burden of organizing the Fourth International Conference on Atomic Masses and Fundamental Constants An Organizing Committee was appointed with the members J H Sanders Chairman B W Petley Secretary A Horsfield Treasurer P Dean A H Wapstra They were assisted by a Program Committee consisting of A H Wapstra Chairman E R Cohen A Horsfield

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