Name	Answer Key	Prif	Date
			- Section

#### Accelerated Chemistry: Average Atomic Mass Worksheet

Calculate the average atomic mass for each element based on the natural abundance of its isotopes.

 Find the average atomic mass for U if 7.5% of U atoms are "U with a mass of 6.0151223 amu and 92.5% are "U with a mass of 7.0160041 amu.

 Find the average atomic mass for B if 19.9% of B atoms are <sup>10</sup>B with a mass of 10.0129371 amu and 80.1% are <sup>13</sup>B with a mass of 11.0093055 amu.

 Find the average atomic mass for CI if 75,78% of CI atoms are <sup>13</sup>CI with a mass of 34.96885271 amu and 24,22% are <sup>37</sup>CI with a mass of 36,96590260 amu.

 Find the average atomic mass for Mg if 78.99% of Mg atoms are <sup>24</sup>Mg with a mass of 23.9850419 amu, 10.00% are <sup>25</sup>Mg with a mass of 24.9858370 amu, and 11.01% are <sup>26</sup>Mg with a mass of 25.9825930 amu.

5. There are 2 isotopes of copper that occur naturally: "Cu and "Cu. The "Cu atoms have a mass of 62.929601 amu and the "Cu atoms have a mass of 64.927794 amu. What is the percent natural abundance for each isotope? (3Cu = 69.15% "Cu = 30.85% (3.546 amu = (62.929601 amu)x + (64.927794 amu)(1-x)

the 2 isotope abundances must equal all of the copper so Let (3Cu = x and (5Cu = 1-x . Now isolate and solve for x, (3.546 amu = (62.929601 amu)x + (64.927794 amu) - (64.927794 amu)x (3.546 amu - 64.927794 amu) = (62.929601 amu) x + (64.927794 amu) x = -1.99893 = 0.6915 = Cu; 65Cu = 1-0.6916 = 0.3085

# **Average Atomic Mass Worksheet Answer Key**

**Ensheng Dong** 

**Average Atomic Mass Worksheet Answer Key:** 

The book delves into Average Atomic Mass Worksheet Answer Key. Average Atomic Mass Worksheet Answer Key is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Average Atomic Mass Worksheet Answer Key, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
  - Chapter 1: Introduction to Average Atomic Mass Worksheet Answer Key
  - o Chapter 2: Essential Elements of Average Atomic Mass Worksheet Answer Key
  - o Chapter 3: Average Atomic Mass Worksheet Answer Key in Everyday Life
  - Chapter 4: Average Atomic Mass Worksheet Answer Key in Specific Contexts
  - ∘ Chapter 5: Conclusion
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- 3. In chapter 2, the author will delve into the foundational concepts of Average Atomic Mass Worksheet Answer Key. This chapter will elucidate the essential principles that must be understood to grasp Average Atomic Mass Worksheet Answer Key in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Average Atomic Mass Worksheet Answer Key in daily life. This chapter will showcase real-world examples of how Average Atomic Mass Worksheet Answer Key can be effectively utilized in everyday scenarios.
- 5. In chapter 4, the author will scrutinize the relevance of Average Atomic Mass Worksheet Answer Key in specific contexts. This chapter will explore how Average Atomic Mass Worksheet Answer Key is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, the author will draw a conclusion about Average Atomic Mass Worksheet Answer Key. The final chapter will summarize the key points that have been discussed throughout the book.

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  - This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Average Atomic Mass Worksheet Answer Key.

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