

WACE CHEMISTRY NOTES 2016

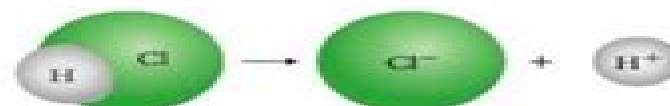
Intro

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Acid

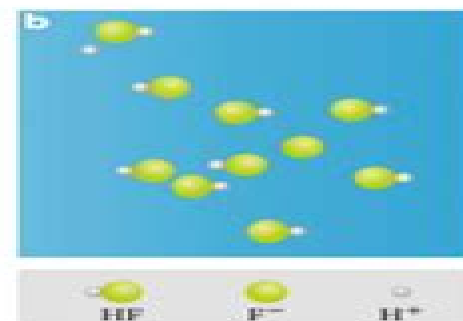
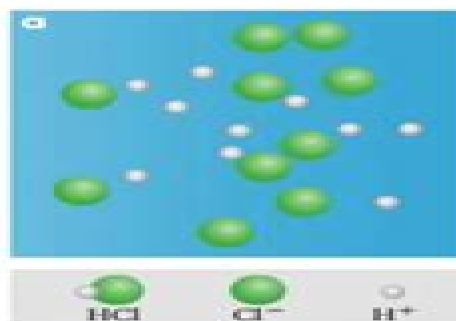
- Produces hydrogen ions
- Or specifically hydronium ions (H_3O^+) in aqueous solution
- Reaction with acid + water \Rightarrow ions are formed \Rightarrow ionisation reaction
- Strong acids \Rightarrow all acid molecules ionise \Rightarrow hydrochloric acid (HCl), Nitric Acid (HNO_3), sulfuric acid (H_2SO_4)

Figure C2.2 ►
Strong acid ionisation:
 $\text{HCl} \rightarrow \text{Cl}^- + \text{H}^+$



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Figure C2.3 ►
Comparison of how
strong acid (HCl) and
weak acid (HF) molecules
ionise in water. a) All of the
HCl molecules are ionised.
b) Only some HF
molecules are ionised.



Base

- Produces hydroxide ion in solution ie. Soluble salts with oxide (O^{2-}) or hydroxide (OH^-) ion
- Soluble base \Rightarrow alkali
- Strong bases always dissociate in water to form separate ions \Rightarrow dissociation reaction

Dot Point Wace Chemistry

**Jack Anderson, Georgia Anderson, Tyler
Clement, Aryan Puri, Rabea Zeynali**

Dot Point Wace Chemistry:

Dot Point WACE Chemistry Units 2A And 2B Marilyn Schell,Margaret Hogan,Peter Wong,2009-01-01 **The American Bookseller** ,1891 **The Nation** ,1877 *Dot Point IB Chemistry AHL* Marilyn Schell,Margaret Hogan,2010

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Dot Point Wace Chemistry Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has are more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Dot Point Wace Chemistry**, a literary masterpiece that delves deep in to the significance of words and their effect on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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