

CHEMISTRY PAPER – I

(Science Pre-Engineering & Pre-Medical Groups)

SECTION 'A'

(MULTIPLE CHOICE QUESTIONS) – (M.C.Qs.)

Time: 20 minutes

(Marks : 17)

Max. Marks: 17

According to New Book

NOTE:

- i) This section consists of 17 part questions and all are to be answered. Each question carries one mark.
 ii) Do not copy the part questions in your answerbook. Write only the answer in full against the proper number of the question and its part.
 iii) The use of scientific calculator is allowed

1. Choose the correct answer for each from the given options:

- i) Generally actual yield is:
 * greater than theoretical yield * **less than theoretical yield**
 * equal to the theoretical yield * sometimes greater than and sometimes less than theoretical yield
- ii) Maximum number of molecules present in the following sample of gas:
 * 100g O₂ * **100g CH₄** * 100g CO₂ * 100g Cl₂
- iii) This crystal system has all sides equal and all angles equal to 90°:
 * Triclinic * Monoclinic * **Cubic** * Hexagonal
- iv) This contains the same no. of atoms as 12g of Mg:
 * 12g C * **20 g Ca** * 24 g C * 12 g Na
- v) The temperature at which two allotropic forms coexist in equilibrium is:
 * Fusion temperature * **Melting point**
 * Critical temperature * Transition temperature
- vi) The volume occupied by 3.01×10^{23} molecules of NH₃ at STP is:
 * 22.4 dm³ * 5.6 dm³ * **11.2 dm³** * 5.4 dm³
- vii) This is not isoelectronic pair:
 * Na⁺ & Ne * Na⁺ & F⁻ * **Na & Ca** * Na⁺ & Mg⁺²
- viii) Which rule or the principle is violated, in the given electronic configuration:
 $sO = 1s^2, 2s^2 2p_x^2 2p_y^2 2p_z^0$
 * Pauli's Exclusion Principle * **(n+l) Rule**
 * Aufbau Principal * **Hund's Rule**
- ix) Which quantum number describes the shape of a orbital:
 * Principle quantum number * **Azimuthal quantum number**
 * Spin quantum number * Magnetic quantum number
- x) This possesses weakest London dispersion forces:
 * Cl₂ * **F₂** * Br₂ * I₂
- xi) Quantum no values for 3s orbital are:
 * n=0, l=1 * n=1, l=0 * n=3, l=1 * **n=3, l=0**
- xii) A colloidal solution of liquid into liquid is known as:
 * Gel * Foam * Aerosol * **Emulsion**
- xiii) Liquids can form convex meniscus in a narrow glass tube when:
 * **Cohesive forces are stronger than adhesive** * Adhesive forces are stronger than cohesive
 * Cohesive forces and adhesive are equal in strength * No hydrogen bonding is there
- xiv) Line spectrum serves as the finger print for the identification of:
 * Electron * Atom * **Element** * Proton
- xv) Hydrogen bond is not found in:
 * H₂O * NH₃ * **CH₄** * HF
- xvi) The diffusion rate of C₂H₆ & CO₂ are same because:
 * Both are polyatomic gases * Both are denser than air
 * **Both have same molecular Mass** * Both contain C atom
- xvii) According to Raoult's Law, the relative lowering of vapor pressure depends on:
 * Mole fraction of solvent * Molarity
 * **Mole fraction of solute** * Molality

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 - The Rise of Digital Reading Dowonload Solved Notes Of Chemistry For Hsc Part
 - Advantages of eBooks Over Traditional Books
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 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
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 - Popular eBook Platforms
 - Features to Look for in an Dowonload Solved Notes Of Chemistry For Hsc Part
 - User-Friendly Interface
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 - Minimizing Distractions
 - Managing Screen Time
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 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

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