

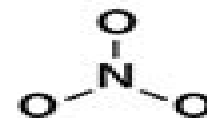
How to Draw Lewis Structure

For Nitrate ion (NO_3^-)

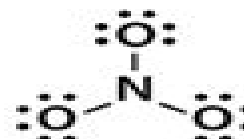
- 1 Determine the number of valence electrons



- 2 Write the skeleton structure



- 3 Fill in the valance electrons



- 4 Transfer electron lone pairs

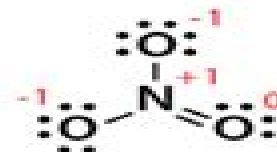


- 5 Determine the formal charge

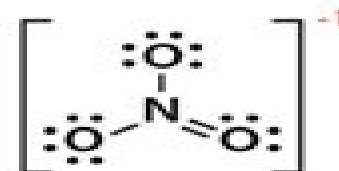
O with single bonds: $6 - 6 - 2/2 = -1$

O with a double bond: $6 - 4 - 4/2 = 0$

N: $5 - 0 - 8/2 = +1$



- 6 Draw a bracket and put the charge



Drawing Electron Dot Diagrams

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