

Quiz & Worksheet - Double Displacement Reaction

<http://study.com/academy/practice/quiz-worksheet-double-displacement-reaction.html>

1. What type of reaction is the following?



☐ combustion

☐ gas formation

☐ precipitation

☐ neutralization

2. What are the ions and the correct charges for the following compound:



☐ Pb^{2+} Cl^{2-}

☐ Pb^{2-} Cl^{2-}

☐ Pb^{+} Cl^{2-}

☐ Pb^{2+} Cl^{-}

3. What are the correct products for the following reaction? (also include the correct coefficient for a balanced reaction)



☐ $\text{Ag}_2\text{SO}_4 + \text{Na}_2\text{NO}_3$

☐ $\text{Ag}_2\text{SO}_4 + \text{NaNO}_3$

☐ $\text{Ag}_2\text{SO}_4 + 2\text{NaNO}_3$

☐ $\text{AgSO}_4 + \text{NaNO}_3$

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Double Replacement Reaction And Solubility Lab Answer Key

Valentine Korah



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Steve Russo, Mike Silver, 2002 Steve Russo and Mike Silver turn chemistry into a memorable story that engages readers and provides the context they need to understand and remember core concepts The book builds interesting applications and well designed illustrations into the narrative to get and hold attention then builds confidence with integrated active learning activities Readers make the connections between concepts and the problem solving techniques they need to master as they read The new edition strengthens this conceptual approach and presents additional quantitative techniques in key areas Readers will find enhanced support for quantitative problem solving and more challenging questions at the end of each chapter in addition to the wealth of technology based support on The Chemistry Place™ Special Edition and on The Chemistry of Life CD ROM For college instructors and students Essentials of Introductory Chemistry Russo Steve Silver Michael, Steve Russo, 2001-12 Introductory Chemistry Third Edition helps readers master the quantitative skills and conceptual understanding they need to gain a deep understanding of chemistry Unlike other books on the market that emphasize rote memory of problem solving algorithms Introductory Chemistry takes a conceptual approach with the idea that focusing on the concepts behind chemical equations helps readers become more proficient problem solvers What Is Chemistry The Numerical Side of Chemistry The Evolution of Atomic Theory The Modern Model of the Atom 1 Chemical Bonding and Nomenclature The Shape of Molecules Chemical Reactions Stoichiometry and the Mole The Transfer of Electrons from One Atom to Another in a Chemical Reaction Intermolecular Forces and the Phases of Matter What If There Were No Intermolecular Forces The Ideal Gas Solutions When Reactants Turn into Products Chemical Equilibrium Electrolytes Acids and Bases For all readers interested in introductory chemistry **Energy Research Abstracts** ,1985

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