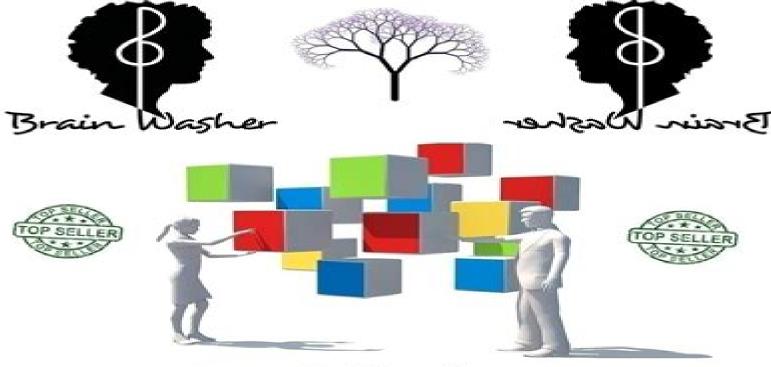
# Data Structures In C

Beginner's Easy Guide. 2014



Harry H. Chaudhary.

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web we know that the mass number a number of protons the number of neutrons and therefore the number of protons is equal to p 35 18 17 and therefore the element is cl the number of protons from electrons for a neutral atom the number of protons and the number of electrons are equal this is what makes the atom charge free

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web together the number of protons and the number of neutrons determine an element s mass number mass number protons neutrons if you want to calculate how many neutrons an atom has you can simply subtract the number of protons or atomic number from the mass number a property closely related to an atom s mass number is its atomic

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web jun 2 2019 number of protons atomic number of the element number of electrons number of protons number of neutrons mass number atomic number find the number of protons each element is defined by the number of protons found in each of its atoms no matter how many electrons or neutrons an atom has the element is defined

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