

ite:	

COUNTING ATOMS

 Chemical: 	formulas can:	have three	components.
-------------------------------	---------------	------------	-------------

a):	The	which are represented by a single uppercase letter, or represented by an
	uppercase letter follow	d by a lowercase letter. This tells you the types of elements in the compound.

- b) The ______ which are numbers that are found on the lower _____ hand side of each element symbol. This tells you the number of _____ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only atom of this element in the molecule.
- c) The _____ which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be _____ by the _____ on the outside of the bracket found on the lower _____ hand side of that bracket.
- 2. Record the number of each atom in each molecule, then record the total number of atoms in the molecule:

I) NaOH	
Element	# of Atoms
Total	
4) Li ₂ SO ₄	70
Element	# of Atoms
Total	
7) H,COCH,	
Element	# of Atoms
Total	
10) NH,C,H,4	O ₂
Element	# of Atoms
Total	
13) Ca ₂ (PO ₄)	
Element	# of Atoms
WALLES THE STATE OF THE STATE O	n or receive

Total

of Atoms
of Atoms

Element	# of Atoms
7.4	
Total	
8) Mg(OH) ₂	
Element	M of Atoms

Element	# of Atoms
Total	

11) NaC ₂ H ₃ O	
Element	# of Atoms
Total	

14) Fe ₂ (CO ₃),	122
Element	# of Atoms
Total	
Lotal	

Element	# of Atoms
	9

6) CaClO,	
Element	# of Atoms
	3
Total	

9) AI(OH),	
Element	# of Atoms
	5
Total	

12) (NH ₄),PO ₄	
Element	# of Atoms
	3
Total	

15) Al ₂ (SO ₃),	123
Element	# of Atoms
Total	

Counting Atoms In Compounds Answer

Pamela Mills, Amina El-Ashmawy

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