

CHEM 120N | EDAPT COVALENT BONDING QUIZ WITH ANSWERS | 2024

Introduction to General Organic and Biological Chemistry with Lab-Mueller > Unit 2

The compound nitrogen trifluoride would have what chemical formula?

N_3F

$(\text{NF})_3$

NF_3

NF

Which of the pairs of atoms below would be expected to make covalent bonds? Select all that apply:

Calcium and nitrogen

Sodium and bromine

Hydrogen and oxygen

Carbon and chlorine

In a covalent bond, electrons are:

shared

destroyed

taken

transferred

Which compounds would you expect to contain one or more covalent bonds

	Covalent Bonds	No Covalent Bonds
	<div><div>H_2O</div><div>S_8</div><div>N_2O_3</div></div> <div><div>S_2F_{10}</div><div>CBr_4</div></div>	<div><div>Ca_3P_2</div><div>CaF_2</div><div>MgO</div></div>

Covalent Bonding Quiz Answers

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