

11**CHEMICAL REACTIONS****Chapter Test A****A. Matching**

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

Column A

- _____ 1. coefficient
- _____ 2. spectator ion
- _____ 3. combustion reaction
- _____ 4. reactant
- _____ 5. skeleton equation
- _____ 6. balanced equation
- _____ 7. activity series
- _____ 8. product
- _____ 9. decomposition reaction
- _____ 10. single-replacement reaction

Column B

- a. a chemical equation that does not indicate relative amounts of reactants and products
- b. a starting substance in a chemical reaction
- c. a list of metals in order of decreasing reactivity
- d. a reaction in which a single compound is broken down into simpler substances
- e. a whole number that appears before a formula in an equation
- f. a new substance formed in a chemical reaction
- g. a particle not directly involved in a chemical reaction
- h. an equation in which each side has the same number of atoms of each element
- i. a reaction in which the atoms of one element replace the atoms of a second element
- j. a reaction in which oxygen reacts with another substance, often producing heat or light

B. Multiple Choice

Choose the best answer and write its letter on the line.

- _____ 11. The symbol Δ in a chemical equation means:
 - a. **heat is supplied to the reaction.**
 - b. a catalyst is needed.
 - c. yields.
 - d. precipitate.
- _____ 12. A catalyst is:
 - a. a solid product of a reaction.
 - b. one of the reactants in a single-replacement reaction.
 - c. **not used up in a reaction.**
 - d. the product of a combustion reaction.

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Amin Saikal,Amitav Acharya



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