

CHAPTER **11** STUDY GUIDE

# Stoichiometry

## Section 11.1 What is stoichiometry?

In your textbook, read about stoichiometry and the balanced equation.

For each statement below, write *true* or *false*.

- \_\_\_\_\_ 1. The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry.
- \_\_\_\_\_ 2. Stoichiometry is based on the law of conservation of mass.
- \_\_\_\_\_ 3. In any chemical reaction, the mass of the products is less than the mass of the reactants.
- \_\_\_\_\_ 4. The coefficients in a chemical equation represent not only the number of individual particles but also the number of moles of particles.
- \_\_\_\_\_ 5. The mass of each reactant and product is related to its coefficient in the balanced chemical equation for the reaction by its molar mass.

Complete the table below, using information represented in the chemical equation for the combustion of methanol, an alcohol.

methanol + oxygen → carbon dioxide + water



Substance	Molar Mass (g/mol)	Number of Molecules	Number of Moles (mol)	Mass (g)
6. Methanol	32.05			
7. Oxygen gas	32.00			
8. Carbon dioxide	44.01			
9. Water	18.02			

10. What are the reactants? \_\_\_\_\_
11. What are the products? \_\_\_\_\_
12. What is the total mass of the reactants? \_\_\_\_\_
13. What is the total mass of the products? \_\_\_\_\_
14. How do the total masses of the reactants and products compare? \_\_\_\_\_

# Chapter 11 Study Guide For Content Mastery Section 11 1 Measuring Matter Answers

**L Manion**



## Chapter 11 Study Guide For Content Mastery Section 11 1 Measuring Matter Answers:

Matter and Energy ,2005      **Prentice Hall Chemistry** ,2000      *Research in Education* ,1971      **Resources in Education** ,1994      **Glencoe Science** McGraw-Hill Staff,2001-08      **North Carolina Education** ,1918      *Ecology: Teacher's ed* ,2005      Onward ,1900      A New English Dictionary on Historical Principles: part 1. C-Comm (1893) James Augustus Henry Murray,1893      *The Saturday Review of Politics, Literature, Science and Art* ,1888      East European Accessions List Library of Congress. Processing Department,1956      *East European Accessions Index* ,1956      **East European Accessions List** ,1956-07      Textile Manufacturer ,1940      **Effective Teaching Methods** Gary D. Borich,1996

This text presents practical yet realistic research based teaching practices that have proven effective for engaging students in the learning process managing their classroom and increasing student achievement It not only tells what to do to obtain these results but shows how to obtain them The text explains how individual differences and learner diversity affect student learning needs and classroom management It also explores teaching strategies in depth explaining how to use both direct instructional methods and indirect methods A self report survey instrument is included to help teachers measure concerns about themselves the teaching task and their impact on students      *Christian Advocate and Journal and Zion's Herald* ,1903

Educational Times ,1886      **Walford's Concise Guide to Reference Material** Albert John Walford,1992 This is a shortened version of the three volume Walford's Guide to Reference Material 5th edition Volume 1 Science and Technology 1989 Volume 2 Social and historical sciences philosophy and religion 1990 and Volume 3 Generalia language and literature the arts 1991 There are more than 3 000 entries forming an updated compilation of what are considered to be the basic items in the main volumes plus some more recent material up to April 1992      *Physics Briefs* ,1979      **The Oxford English Dictionary** ,1989 In addition to current definitions provides an historical treatment to words and idioms included

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## **Chapter 11 Study Guide For Content Mastery Section 11.1 Measuring Matter Answers**

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