

CHAPTER 12 REVIEW*Solutions***SECTION 3****SHORT ANSWER** Answer the following questions in the space provided.**1.** Describe the errors made by the following students in making molar solutions.

- a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L. of water to the solid.

- b. Mary needs a 0.02 M solution of NaNO_3 . She calculates that she needs 2.00 g of NaNO_3 for 0.02 mol. She puts this solid into a 1.00 L. volumetric flask and fills the flask to the 1.00 L. mark.

PROBLEMS Write the answer on the line to the left. Show all of your work in the space provided.

- 2.** _____ What is the molarity of a solution made by dissolving 2.0 mol of solute in 6.0 L. of solvent?

- 3.** _____ CH_3OH is soluble in water. What is the molality of a solution made by dissolving 8.0 g of CH_3OH in 250. g of water?

Chapter 12 Review Solutions Section 3 Answers

Sinha



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