

Stoichiometry

- Mass-Mass Example #2:



- How many grams of nitrogen are needed to produce the 38.5 g of NH_3 produced in the previous example?

38.5 g NH_3	1 mol NH_3	1 mol N_2	28.0 g N_2
	17.0 g NH_3	2 mol NH_3	1 mol N_2
= 31.7 g N_2			

Chapter 12 Stoichiometry 12 1 Arithmetic Equations 17

**Charles William Keenan, Donald C.
Kleinfelter, Jesse Hermon Wood**



Chapter 12 Stoichiometry 12.1 Arithmetic Equations 17:

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