

THERMOCHEMISTRY-HEAT AND CHEMICAL CHANGE
Chapter 11, 19.3 and 19.4

Name **KEY**

Date _____ Pd _____

Write the letter of the best answer in the blank.

- _____ 1. Thermochemistry is the study of:
a. reaction rates
b. heat changes during reactions
c. combustion reactions
d. stoichiometry relationships
- _____ 2. A sample of glass that has a mass of 6.0 g gives off 12 J of heat. If the temperature of the sample changes by 4.0°C during this change, what is the specific heat of the glass?
a. 2.0 J/(g x °C)
c. 0.50 J/(g x °C)
b. 1.1 J/(g x °C)
d. 0.45 J/(g x °C)
- _____ 3. The accurate and precise measurement of the heat change for chemical and physical properties is called:
a. calorimetry
b. stoichiometry
c. thermochemistry
d. thermometry
- _____ 4. Two water solutions that have a density of 1.00 g/mL are mixed, and a reaction occurs. The temperature rises from 25.0°C to 27.0°C. Given that the heat capacity is 4.18 J/(g x °C) and that the total mass of the water is 75.0 g, what is the heat change for the reaction?
a. 6.27 J
b. 125 J
c. 314 J
d. 627 J
- _____ 5. If 4.0 mol of a substance releases 12 kJ when it decomposes, what is the ΔH for the process, in terms of kJ per mole of reactant?
a. 3.0 kJ
b. -3.0 kJ
c. 0.33 kJ
d. -0.33 kJ
- _____ 6. The heat absorbed by 1 mol of a substance in melting from a solid to a liquid at constant temperature is symbolized by which of these symbols?
a. ΔH_{m}
b. ΔH_{solid}
c. ΔH_{vap}
d. ΔH_{soln}
7. When a chemical equation is reversed the sign of ΔH _____. 7. **is reversed**
8. The value of ΔH_f° of any free element in the standard state is set at _____. 8. **0.0 kJ/mol**
9. The standard heat of reaction is equal to ΔH_f° of the products _____ ΔH_f° of the reactants. 9. **minus**

Answer the following in the space provided.

10. State whether the following physical and chemical changes are endothermic or exothermic.
- | | | | |
|-----------------|--------------------|---------------|--------------------|
| a. melting | endothermic | d. fusion | endothermic |
| b. vaporization | endothermic | e. freezing | exothermic |
| c. condensation | exothermic | f. combustion | exothermic |

Chapter 115answer Key

Didier Musso



Chapter 11answer Key:

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1. Understanding the eBook Chapter 115answer Key
 - The Rise of Digital Reading Chapter 115answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 115answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 115answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 115answer Key
 - Personalized Recommendations

- Chapter 115answer Key User Reviews and Ratings
- Chapter 115answer Key and Bestseller Lists
- 5. Accessing Chapter 115answer Key Free and Paid eBooks
 - Chapter 115answer Key Public Domain eBooks
 - Chapter 115answer Key eBook Subscription Services
 - Chapter 115answer Key Budget-Friendly Options
- 6. Navigating Chapter 115answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 115answer Key Compatibility with Devices
 - Chapter 115answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 115answer Key
 - Highlighting and Note-Taking Chapter 115answer Key
 - Interactive Elements Chapter 115answer Key
- 8. Staying Engaged with Chapter 115answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 115answer Key
- 9. Balancing eBooks and Physical Books Chapter 115answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 115answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 115answer Key
 - Setting Reading Goals Chapter 115answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 115answer Key
 - Fact-Checking eBook Content of Chapter 115answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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