

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	Aluminum
_____	14-18
_____	Transition?

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It exists in the periodic table.
- _____ 20. It has no line periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are strong oxidizing agents.
- _____ 23. Group 18 elements.

Column B

- a. noble
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in bold; change the underlined word or phrase to make it true.

- _____ 24. There are two representative elements of aluminum.
- _____ 25. Most elements have fewer of the elements in the periodic table are noble.
- _____ 26. Group 14 elements except for hydrogen and boron in the alkali metals.
- _____ 27. Group 18 elements are the alkali metal noble.
- _____ 28. Group 14 elements are highly reactive transition elements in hydrogen.
- _____ 29. Group 14 elements are very reactive elements found in transition metals.
- _____ 30. Alkali metals have properties of both metals and non-metals.

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