

# CBSE Class 12 Chemistry Chapter 13 Amines Worksheet – Set 2

Q1. How many isomeric amines with the formula C<sub>2</sub>H<sub>0</sub>N contain a benzene ring?

- (a) Five
- (b) Four
- (c) Three
- (d) None of the above

Q2. The exidation of aniline with K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> / H<sub>2</sub>SO<sub>4</sub> produces

- (a) Benzoic acid
- (b) p- Benzo quinone
- (c ) p- Nitro phenol
- (d) None of the above

Q3. The electrolytic reduction of nitro benzene in a strongly acidic medium produces

- (a) Aniline
- (b) Phenyl hydroxy amine
- (c) p- amino phenol
- (d) None of the above

Q4. In the nitration of benzene with concentrated nitric acid and concentrated sulphuric acid, the electrophile is

- (a) NO,"
- (b) NO.
- (c) NO.
- (d) None of the above

Q5. An organic amino compound reacts with aqueous nitrous acid at low temperatures to produce an oily nitrosamine. The compound is

- (a) (C<sub>2</sub>H<sub>5</sub>)<sub>5</sub>N
- (b) (C<sub>2</sub>H<sub>2</sub>)<sub>2</sub>NH
- (c) C<sub>2</sub>H<sub>2</sub>NH<sub>2</sub>
- (d) None of the above

Q6. Write the IUPAC and common name of the following compound.

Q7. Draw the structure of m- toluidine.

Q8. Identify A, B and C in the reaction mentioned below.

$$CH_3Br \xrightarrow{ECN} A \xrightarrow{LiACH_4} B \xrightarrow{RNO_2} C$$

Q9. Convert nitrobenzene to aniline.

Q10. Why does methyl amine have a lower boiling point than methanol?

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**Thomas Kerchever Arnold** 

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