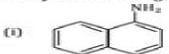
Chemistry

(www.tiwariacademy.com) (Chapter 13) (Amines) (Intext Questions) (Class 12)

Ouestion 13.1:

Classify the following amines as primary, secondary or tertiary:



(iiii) (C2H5)2CHNH2 Answer 13.1:

Primary: (i) and (iii) Secondary: (iv) Tertiary: (iii)

Question 13.2:

(i) Write structures of different isomeric amines corresponding to the molecular formula, C₄H₁₁N (ii) Write IUPAC names of all the isomers.

(iii) What type of isomerism is exhibited by different pairs of amines?

_Answer 13.2:

(i), (ii) The structures and their IUPAC names of different isomeric amines corresponding to the molecular formula, CaH11N are given below:

N-Mathylpropan-2-amine (2°) (iii) The pairs (a) and (b) and (e) and (g) exhibit position isomerism.

The pairs (a) and (c); (a) and (d); (b) and (c); (b) and (d) exhibit chain isomerism. The pairs (e) and (f) and (f) and (g) exhibit metamerism.

All primary amines exhibit functional isomerism with secondary and tertiary amines and vice-versa.

Question 13.3:

How will you convert?

(i) Benzene into aniline

(ii) Benzene into N, N-dimethylaniline

(iii) Cl-(CH₂)₄-Cl into hexan-1, 6-diamine?

 $N(CH_2)_2$

CH₃- CH- CH₂-NH₂

CH₃—C—NH₂

2-Methylpropan-2-amine (1°) CH3-CH2-NH-CH2-CH3

N-Ethylethanamine (2°).

N. N. Dimethylethanamine (3°)

CH₃ — CH₂ — CH₃

Butan-2-amine (1°)

(iv) (C2Hs)2N

(b)

(d)

(n)

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