

## Chemistry Worksheet Class 11 on Chapter 12 Organic Chemistry – Some Basic Principles & Techniques – Set 1

**Q1.** Which of the following carbocation is most stable?

- (a)  $(\text{CH}_3)_2 \text{CH}^+$
- (b)  $\text{Ph}_3 \text{C}^+$
- (c)  $\text{CH}_3 \text{CH}_2^+$
- (d) None of the above

**Q2.** The molecule in which the distance between the neighbouring carbon atom is least is an \_\_\_\_\_.

- (a) Alkane
- (b) Alkene
- (c) Alkyne
- (d) All of them have equal distance

**Q3.** What is the general formula of cycloalkanes?

- (a)  $\text{C}_n\text{H}_{2n}$
- (b)  $\text{C}_n\text{H}_{2n+1}$
- (c)  $\text{C}_n\text{H}_{2n+2}$
- (d) None of the above

**Q4.** Which of the following is not a cyclic compound?

- (a) Neopentane
- (b) Phenol
- (c) Pyrrole
- (d) None of the above

**Q5.** Which of the following hybridisation has maximum s characteristics?

- (a)  $sp$
- (b)  $sp^2$
- (c)  $sp^3$
- (d) All of them have the same s characteristic

**Q6.** How many sigma and pi bonds are present in 1,2- propadiene?

**Q7.** What is homologous series?

**Q8.** How many isomers are possible for monosubstituted and disubstituted benzene?

# Chapter 12 Basic Review Worksheet Chemistry Answers

**Holt Rinehart & Winston, Holt,  
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**Children's Books in Print** R R Bowker Publishing, Bowker, 1999-12      **Science Insights** ,1999      **Linne & Ringsrud's Clinical Laboratory Science - E-Book** Mary Louise Turgeon, 2015-02-10 Using a discipline by discipline approach Linne perforated pages make it easy for students to remove for use in the lab and for assignment of review questions as homework NEW Instrumentation updates show new technology being used in the lab NEW Additional key terms in each chapter cover need to know terminology NEW Additional tables and figures in each chapter clarify clinical lab science concepts      **Biology** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004      **Integrating Green and Sustainable Chemistry Principles into Education** Andrew P. Dicks, Loyd D. Bastin, 2019-07-19 Integrating Green and Sustainable Chemistry Principles into Education draws on the knowledge and experience of scientists and educators already working on how to encourage green chemistry integration in their teaching both within and outside of academia It highlights current developments in the field and outlines real examples of green chemistry education in practice reviewing initiatives and approaches that have already proven effective By considering both current successes and existing barriers that must be overcome to ensure sustainability becomes part of the fabric of chemistry education the book's authors hope to drive collaboration between disciplines and help lay the foundations for a sustainable future Draws on the knowledge and expertise of scientists and educators already working to encourage green chemistry integration in their teaching both within and outside of academia Highlights current developments in the field and outlines real examples of green chemistry education in practice reviewing initiatives and approaches that have already proven effective Considers both current successes and existing barriers that must be overcome to ensure sustainability      **Characterizing Pedagogical Flow** W.H. Schmidt, 2005-12-17 Characterizing Pedagogical Flow presents conclusions from a multi disciplinary multi national research project blending quantitative and qualitative approaches through a discourse methodology The work produced portraits of mathematics and science education that were dramatically different for each of the countries involved France Japan Norway Spain Switzerland and the United States To explain these differences it is proposed that the interaction of curriculum and pedagogy is culturally unique and yields classroom learning experiences that are qualitatively different from country to country This idea has profound implications for how international education research is interpreted      *Holt Chemistry* Ralph Thomas Myers, 2004      Addison-Wesley Science Insights ,1996      **Health** Frances Sienkiewicz Sizer, 2000 This student textbook for grades 9 12 covers emotional and physical health topics along with drug abuse information human reproduction general fitness and nutrition and the environment and health impact each other      **Curriculum Review** ,1986      **Te HS&T J** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02      **Programmed Learning and Individually Paced Instruction** Carl H. Hendershot, 1973      **Te HS&T 2007 Shrt Crs M** Holt Rinehart & Winston, 2007      R.R. Bowker's Software for Schools ,1987      Exploring Earth and Space Michael DiSpezio, 1995 A textbook exploring such

aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter *Glencoe Science* McGraw-Hill Staff,2001-06 *Science Spectrum* Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2003-03 **CAP Today** ,1990 **A Level Chemistry Multiple Choice Questions and Answers (MCQs)** Arshad Iqbal,2020-04-10 Previously published as A Level Chemistry MCQs Multiple Choice Questions and Answers Quiz Tests with Answer Keys by Arshad Iqbal A Level Chemistry Multiple Choice Questions and Answers MCQs A Level Chemistry quizzes practice tests with answer key provides mock tests for competitive exams to solve 1745 MCQs A Level Chemistry MCQs helps with theoretical conceptual and analytical study for self assessment career tests This book can help to learn and practice A Level Chemistry quizzes as a quick study guide for placement test preparation A level Chemistry Multiple Choice Questions and Answers MCQs is a revision guide with a collection of trivia quiz questions and answers on topics Alcohols and esters atomic structure and theory benzene chemical compound carbonyl compounds carboxylic acids acyl compounds chemical bonding chemistry of life electrode potential electrons in atoms enthalpy change equilibrium group IV groups II and VII halogenoalkanes hydrocarbons introduction to organic chemistry ionic equilibria lattice energy moles and equations nitrogen and sulfur organic and nitrogen compounds periodicity polymerization rates of reaction reaction kinetics redox reactions and electrolysis states of matter transition elements to enhance teaching and learning A level Chemistry Quiz Questions and Answers also covers the syllabus of many competitive papers for admission exams of different universities from chemistry textbooks on chapters Alcohols and Esters Multiple Choice Questions 27 MCQs Atomic Structure and Theory Multiple Choice Questions 37 MCQs Benzene Chemical Compound Multiple Choice Questions 41 MCQs Carbonyl Compounds Multiple Choice Questions 29 MCQs Carboxylic Acids and Acyl Compounds Multiple Choice Questions 27 MCQs Chemical Bonding Multiple Choice Questions 213 MCQs Chemistry of Life Multiple Choice Questions 29 MCQs Electrode Potential Multiple Choice Questions 62 MCQs Electrons in Atoms Multiple Choice Questions 53 MCQs Enthalpy Change Multiple Choice Questions 45 MCQs Equilibrium Multiple Choice Questions 50 MCQs Group IV Multiple Choice Questions 53 MCQs Groups II and VII Multiple Choice Questions 180 MCQs Halogenoalkanes Multiple Choice Questions 33 MCQs Hydrocarbons Multiple Choice Questions 53 MCQs Introduction to Organic Chemistry Multiple Choice Questions 52 MCQs Ionic Equilibria Multiple Choice Questions 56 MCQs Lattice Energy Multiple Choice Questions 33 MCQs Moles and Equations Multiple Choice Questions 50 MCQs Nitrogen and Sulfur Multiple Choice Questions 89 MCQs Organic and Nitrogen Compounds Multiple Choice Questions 54 MCQs Periodicity Multiple Choice Questions 202 MCQs Polymerization Multiple Choice Questions 36 MCQs Rates of Reaction Multiple Choice Questions 39 MCQs Reaction Kinetics Multiple Choice Questions 52 MCQs Redox Reactions and Electrolysis Multiple Choice Questions 55 MCQs States of Matter Multiple Choice Questions 66 MCQs Transition Elements Multiple Choice Questions 29 MCQs The chapter Alcohols and Esters MCQs covers topics of introduction to alcohols and alcohols reactions The chapter Atomic Structure and Theory MCQs covers topics of

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