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Assessment

Chapter 12 Practice Test

Chapter: Solutions

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- Agitation prevents settling in a(n)
 - a. alloy.
 - b. homogeneous mixture.
 - c. suspension.
 - d. gaseous mixture.
- All of the following are heterogeneous mixtures except
 - a. whole wheat bread.
 - b. granite.
 - c. tap water.
 - d. an oil-water mixture.
- 3. What is the concentration of a 100, mL aqueous solution that contains 1.00 g KCl (molar mass = 74.55 g/mol)?
 - a. 1.34 M KCl
 - b. 0.134 M KCI
 - c. 0.0134 M KCI
 - d. 0.001 34 M KCl
- 4. To determine the molarity of an HCl solution, you need to know the number of
 - a. grams of HCl in 1 × 10⁶ g of solution.
 - b. moles of HCl dissolved in the total moles of solution.
 - c. moles of HCl in 1 L of solution.
 - d. moles of HCl dissolved in 1 kg of solvent.
- 5. What type of solute-solvent combination is carbon dioxide in water?
 - a. gas-liquid
 - b. liquid-gas
 - e. liquid-liquid
 - d. cannot be determined
- _____ 6. What is the molarity of a solution that contains 0.202 mol KCl (molar mass = 74.55 g/mol) in 7.98 L, of solution?
 - a. 0.0132 M KCI
 - b. 0.0253 M KCI
 - c. 0.459 M KCl
 - d. 1.36 M KCl

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Reza Nazari, Ava Ross

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